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PROJECTOR



# Kodak Carousel 800 Manual

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$$\text{Pb(NO}_3)_2(\text{aq}) + 2\text{HCl(aq)} \rightarrow \text{PbCl}_2(\text{s}) + 2\text{HNO}_3$$

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web answer key to practice problems on net ionic equations molecular  $\text{AgNO}_3(\text{aq}) + \text{KCl(aq)} \rightarrow \text{AgCl(s)} + \text{KNO}_3(\text{aq})$  total ionic  $\text{Ag}^+(\text{aq}) + \text{Cl}^-(\text{aq})$  net ionic  $\text{Ag}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{AgCl(s)}$

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