Prelim Topic 1: The Chemical Earth

1. The living and non-living components of the Earth contain mixtures

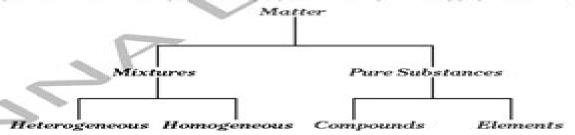
Identify the difference between elements, compounds and mixtures in terms of particle theory.

Particle theory

All matter are made up of tiny particles which are continually moving.

Types of matter

- Pure substance contains fixed compositions, fixed properties and cannot be decomposed by simple physical separation techniques.
 - a. Elements are composed of atoms or molecules
 - Simplest pure substance consisting of one type of atom
 - Cannot decompose
 - Exists as molecules (substances with two or more atoms of the same type chemically combined)
 - b. Compounds are composed of fixed numbers of atoms of different elements
 - Composed of two or more elements chemically bonded together
 - Fixed number of atoms of each component element
 - Decompose into component elements through chemical means
- Mixture (Impure) have various particle types and compositions.
 - a. Homogenous mixtures have uniform composition and properties throughout eg. Air, water
 - b. Heterogeneous mixtures have variable composition and properties throughout eg. Blood, concrete
 - Components of mixture keep their own properties and separated by physical means eg. (filtering)



 Identify that the biosphere, lithosphere, hydrosphere and atmosphere contain examples of mixtures of elements and compounds

Biosphere	 the region on Earth where living organisms are found most living things are composed of cells which contain complex carbon compounds such as carbohydrates, fats and proteins 	Major form	Common elements
		Water	Hydrogen, oxygen
		Calcium phosphate (bones)	Phosphorus, calcium
		Carbohydrates, fats	Carbon, hydrogen, oxygen
		Proteins	Carbon, hydrogen, oxygen, nitrogen, sulphur

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