

Total No. of Questions – 21

Regd.

Total No. of Printed Pages – 2

No.

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Part - III
CHEMISTRY, Paper-II
(English Version)

Time : 3 Hours]

[Max. Marks : 60

Note : Read the following instructions carefully :

- (1) Answer **all** questions of Section – 'A'. Answer any **six** questions in Section – 'B' and any **two** questions in Section – 'C'.
- (2) In Section – 'A', questions from Sr. Nos. 1 to 10 are of "Very short answer type". Each question carries **two** marks. Every answer may be limited to **two** or **three** sentences. Answer all these questions at one place in the same order.
- (3) In Section – 'B', questions from Sr. Nos. 11 to 18 are of "Short answer type". Each question carries **four** marks. Every answer may be limited to **75** words.
- (4) In Section – 'C', questions from Sr. Nos. 19 to 21 are of "Long answer type". Each question carries **eight** marks. Every answer may be limited to **300** words.
- (5) Draw labelled diagrams, wherever necessary for questions in Section – 'B' and 'C'.

SECTION – A**10 × 2 = 20****Note :** Answer **all** the questions :

1. Define Osmotic Pressure.
2. What are antibiotics ? Give example.
3. Write the difference between a soap and a synthetic detergent.
4. State Faraday's first law of electrolysis.
5. Give the composition of the following alloys :
 (a) Brass (b) German Silver
6. List out any two uses of Neon.
7. Write the structure of XeF_4 .
8. Why Zn^{2+} is diamagnetic whereas Cr^{3+} is paramagnetic ?
9. What is biodegradable polymer ? Give one example.
10. Write the names of the monomers of the following polymers :
 (a) Bakelite (b) Nylon - 6,6

Intermediate Chemistry Question Paper 201

Jianjun Gao



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