

**Section 1.1** *continued*

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

19
39
55
87
101
119

Atomic Mass	1.1
Atomic Number	1.2
Electron Configuration	1.3
Chemical Name	1.4
Chemical Symbol	1.5

Put each item in Column A, under the letter of the matching item in Column B.

Column A

- _____ 16. It appears in the periodic table.
- _____ 17. It has no line periodic table.
- _____ 18. Group 1 elements.
- _____ 19. Elements that are strong oxidizing agents.
- _____ 20. Group 8 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ 21. There are two representative elements of elements.
- _____ 22. Most elements have lines in the periodic table for metals.
- _____ 23. Group 1 elements except for hydrogen are metals in the solid state.
- _____ 24. Group 2 elements are the alkaline earth metals.
- _____ 25. Group 17 elements are highly reactive nonmetals known as halogens.
- _____ 26. Group 8 elements are very reactive elements known as noble gases.
- _____ 27. Elements have properties of both metals and nonmetals.

Hydrocarbons Study Guide Answers For Content Mastery

S Ashworth



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