## HALF-LIFE OF RADIOACTIVE ISOTOPES

Name

 How much of a 100.0 g sample of <sup>108</sup>Au is left after 8.10 days if its half-life is 2.70 days?

$$\frac{8.1}{2.7} = 3$$
  $\frac{100}{2} = 50$   $\frac{50}{2} = 12.5$ 

A 50.0 g sample of <sup>16</sup>N decays to 12.5 g in 14.4 seconds. What is its half-life?

7 200

The half-life of 4k is 12.4 hours. How much of a 750 g sample is left after 62.0 hours?

$$\frac{62}{12.4} = 5$$
  $\frac{(\frac{750}{2})}{2} = 1625$   $\frac{(\frac{1875}{2})}{2} = 46.875$   $\frac{46.875}{2} = 23.4$ 

4. What is the half-life of "Tc if a 500 g sample decays to 62.5 g in 639,000 years?

$$(\frac{1}{2}) + 2 = \frac{1}{8}$$

213000 va

 The half-life of <sup>222</sup>Th is 1.4 x 10<sup>10</sup> years. If there are 25.0 g of the sample left after 2.8 x 10<sup>10</sup> years, how many grams were in the original sample? 2.8 -1000

100.0a

 There are 5.0 g of <sup>131</sup> left after 40.35 days. How many grams were in the original sample if its half-life is 8.07 days?

(00 cm

# **Half Life Calculations Worksheet Answers**

Linda L. Lilley, Robert S. Aucker

#### **Half Life Calculations Worksheet Answers:**

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