

2.5

ELEMENTS AND COMPOUNDS

Section Review

Objectives

1. Explain the difference between elements and a compound
2. Distinguish between a mixture and a solution
3. Identify the chemical symbols for elements and write a chemical formula for a compound

Vocabulary

- | | |
|------------|----------------|
| 1. element | 2. compound |
| 3. mixture | 4. homogeneous |

Part A Completion

Fill in the blank with the word or phrase that best completes or explains each statement. Write your answer in the space provided. (10 points)

- | | |
|---|---|
| 1. An element is a substance that cannot be broken down into simpler substances by chemical means. | 6. A compound is a substance formed by the chemical combination of two or more elements in a fixed ratio. |
| 2. A mixture is a combination of two or more substances in which the components retain their individual properties. | 7. A solution is a homogeneous mixture of two or more substances. |
| 3. A chemical formula is a symbolic representation of a chemical compound. | 8. A chemical reaction is a process in which one or more substances are converted into one or more new substances. |
| 4. A chemical reaction is a process in which one or more substances are converted into one or more new substances. | 9. A chemical reaction is a process in which one or more substances are converted into one or more new substances. |
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