

PERIODIC TABLE PRESENTATION & CLOZE NOTES

PERIODIC TABLE ELEMENTS

What is the Periodic Table?
The periodic table is a way to display all the known elements.

What are Elements?
An element is a type of matter that cannot be broken down into simpler substances.

Examples of Elements
Gold, Silver, Iron, Copper, etc.

What are element Symbols?
It is the chemical symbol for Hydrogen.

How do elements get their Symbols?

- Symbols come from the names of the elements, many from the Latin name of the element.
- The chemical symbol can be one or two letters.
- The first letter is capital and the second is lower case.

Let's review Atoms.
An atom is the smallest part of an element that has all the properties of the element.

What can you learn about atoms from the Periodic Table?

- To find the number of protons in one atom of an element, look at the atomic number.

What is the Atomic Number?
Atomic number represents the number of protons.

Period Example

- Sodium, magnesium, aluminium, silicon, sulfur, phosphorus, chlorine, and argon are found in the same period in the periodic table.
- They are in the same row.
- They all have the same number of electron shells.

Group Example

- Sodium, lithium, potassium, rubidium, and francium are found in the same group in the periodic table.
- They have similar properties because they have the same number of valence electrons.
- These elements are in the same column in the periodic table.

What is the Atomic Mass?
Atomic mass is the sum of the number of protons and neutrons in an atom.

Periodic Table Quick Facts

- The word "periodic" means repeating a pattern.
- Mendeleev observed that elements with similar properties occurred periodically.
- He was the first to group elements in periods (rows) and groups (columns) by their characteristics.
- Elements in the same group have similar properties.
- Elements in the same period have the same number of electron shells.

Element Periodic Table Cloze

Emilie Sanchez

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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