

## CHAPTER 4 REVIEW

**Arrangement of Electrons in Atoms****SECTION 2****SHORT ANSWER** Answer the following questions in the space provided.

1. \_\_\_\_\_ How many quantum numbers are used to describe the properties of electrons in atomic orbitals?  
(a) 1 (c) 3  
(b) 2 (d) 4
2. \_\_\_\_\_ A spherical electron cloud surrounding an atomic nucleus would best represent  
(a) an  $s$ -orbital. (c) a combination of two different  $p$ -orbitals.  
(b) a  $p$ -orbital. (d) a combination of an  $s$  and a  $p$ -orbital.
3. \_\_\_\_\_ How many electrons can an energy level of  $n = 4$  hold?  
(a) 32 (c) 8  
(b) 24 (d) 6
4. \_\_\_\_\_ How many electrons can an energy level of  $n = 2$  hold?  
(a) 32 (c) 8  
(b) 24 (d) 6
5. \_\_\_\_\_ Compared with an electron for which  $n = 2$ , an electron for which  $n = 4$  has more  
(a) spin. (c) energy.  
(b) particle nature. (d) wave nature.
6. \_\_\_\_\_ According to Bohr, which is the point in the figure below where electrons cannot reside?  
(a) point A (c) point C  
(b) point B (d) point D



7. \_\_\_\_\_ According to the quantum theory, point D in the above figure represents:  
(a) the fixed position of an electron.  
(b) the furthest position from the nucleus that an electron can achieve.  
(c) a position where an electron probably exists.  
(d) a position where an electron cannot exist.

# Electrons In Atoms Test Review Answer

**Regents Exambusters**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the 'Regents Exambusters' text.

## **Electrons In Atoms Test Review Answer:**

## **Electrons In Atoms Test Review Answer** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Electrons In Atoms Test Review Answer**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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### **Electrons In Atoms Test Review Answer Introduction**

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