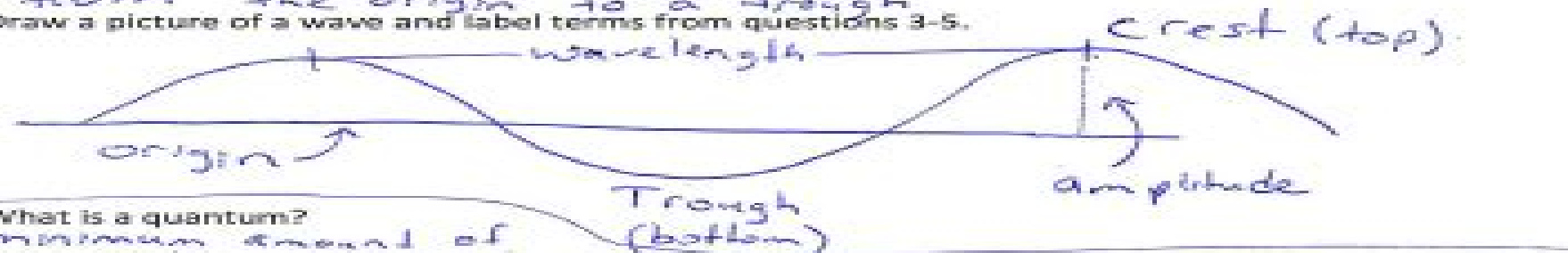


Chapter 5 Study Guide

1. What is electromagnetic radiation? Visible light form of energy that exhibits wavelike behavior as it travels through space. (Energy)
2. What is the atomic emission spectrum? Set of frequencies of the electromagnetic waves emitted by atoms of the element.
3. Define wavelength. Shortest distance between equivalent points on a continuous wave.
4. Define frequency. Number of waves that pass a given point per second
5. Define amplitude. Waves height from the origin to a crest or from the origin to a trough
6. Draw a picture of a wave and label terms from questions 3-5.



7. What is a quantum? minimum amount of energy that can be gained or lost by an atom
8. What does the s-orbital look like? (draw)

 like a ball

9. What does the p-orbital look like? (draw)

 dumbbell shape

10. What does the d-orbital look like? (draw)



11. What is the maximum number of electrons are in the s-orbital?

2 electrons

12. What is the maximum number of electrons are in the p-orbital?

6 electrons

13. What is the maximum number of electrons are in the d-orbital?

10 electrons

14. What is the maximum number of electrons are in the f-orbital?

14 electrons

Electrons In Atoms Study Guide Answer Key

CH Cherryholmes



Electrons In Atoms Study Guide Answer Key:

Electrons In Atoms Study Guide Answer Key Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has become more evident than ever. They have the capability to inspire, provoke, and ignite change. Such could be the essence of the book **Electrons In Atoms Study Guide Answer Key**, a literary masterpiece that delves deep into the significance of words and their affect our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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