

Chapter 1

Introduction to Electronics

Question A.1: Atomic Structure

- (a) The atom is the smallest constituent of matter that retains all the chemical properties of the element.
- (b) The atom is made up of three subatomic particles: electrons, protons, and neutrons.

Question A.2: Materials Used for Electronics

- (a) The materials used for making electronic components are classified into three categories: conductors, semiconductors, and insulators.
- (b) The most commonly used semiconductor material is silicon.
- (c) The most commonly used insulating material is glass.

Question A.3: Current in Conductors

- (a) Current is the flow of electric charge. It is measured in amperes (A).
- (b) Current is produced when there is a potential difference across a conductor.

Question A.4: N-Type and P-Type Semiconductors

- (a) N-type semiconductor is formed by adding a small amount of pentavalent impurity to an intrinsic semiconductor. The impurity atoms have five valence electrons, one of which is free to move, creating free electrons.
- (b) P-type semiconductor is formed by adding a small amount of trivalent impurity to an intrinsic semiconductor. The impurity atoms have three valence electrons, creating holes.

Question A.5: The PN Junction

- (a) The PN junction is formed by joining an N-type semiconductor and a P-type semiconductor. The junction is formed by the diffusion of dopant atoms from the N-type region into the P-type region, creating a depletion region.
- (b) The depletion region is a region where the concentration of free electrons and holes is very low. It acts as an insulator.

Electronic Devices 9th Edition Solution Manual

Daisy Thomas



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