

Practice Problems (Chapter 8): Electron Configuration

CHEM 1A

1. Write complete electron configurations for the following atoms:a) F $1s^2 2s^2 2p^5$

b) Al

c) V

d) Cu

e) Kr

2. Write noble gas electron configurations for the following atoms:a) Cd $[\text{Kr}] 5s^2 4d^{10}$

b) I

c) Ba

d) Rn

3. How many core, outer, and valence electrons do each of the following atoms have?

	Core	Outer	Valence
a) H	0	1	1
b) Si	10	4	4
c) Ti	18	2	4
d) Br	28	7	7
e) Ag	36	1	11

(3d e^- are core e^- for rep. elements)

(an exception)

4. Write orbital notations for the following atoms:

a) O $\frac{1\downarrow}{1s} \frac{1\downarrow}{2s} \frac{1\downarrow}{2p} \frac{1}{2p} \frac{1}{2p}$ $\frac{\quad}{3s}$ $\frac{\quad}{3p} \frac{\quad}{3p} \frac{\quad}{3p}$ $\frac{\quad}{4s}$ $\frac{\quad}{3d} \frac{\quad}{3d} \frac{\quad}{3d}$ b) P $\frac{1\downarrow}{1s} \frac{1\downarrow}{2s} \frac{1\downarrow}{2p} \frac{1\downarrow}{2p} \frac{1\downarrow}{2p} \frac{1\downarrow}{3s} \frac{1}{3p} \frac{1}{3p} \frac{1}{3p}$ $\frac{\quad}{4s}$ $\frac{\quad}{3d} \frac{\quad}{3d} \frac{\quad}{3d}$ c) Cr (an exception) $\frac{1\downarrow}{1s} \frac{1\downarrow}{2s} \frac{1\downarrow}{2p} \frac{1\downarrow}{2p} \frac{1\downarrow}{2p} \frac{1\downarrow}{3s} \frac{1\downarrow}{3p} \frac{1\downarrow}{3p} \frac{1\downarrow}{3p} \frac{1}{4s} \frac{1}{3d} \frac{1}{3d} \frac{1}{3d} \frac{1}{3d} \frac{1}{3d}$ d) Co $\frac{1\downarrow}{1s} \frac{1\downarrow}{2s} \frac{1\downarrow}{2p} \frac{1\downarrow}{2p} \frac{1\downarrow}{2p} \frac{1\downarrow}{3s} \frac{1\downarrow}{3p} \frac{1\downarrow}{3p} \frac{1\downarrow}{3p} \frac{1\downarrow}{4s} \frac{1\downarrow}{3d} \frac{1\downarrow}{3d} \frac{1}{3d} \frac{1}{3d} \frac{1}{3d}$

Electron Configuration Problems

Roman Wölfel



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difficult problems will challenge the student's capacity on finding solutions

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Quantum Mechanics 500 Problems with Solutions Mr. Sanjeev Pandey, 2024-08-16 This book provides 500 carefully selected problems in quantum mechanics each with a detailed solution designed to strengthen conceptual understanding and problem solving skills for students and researchers in physics and chemistry.

Problems in Structural Inorganic Chemistry Wai-Kee Li, Yu-San Cheung, Kendrew Kin Wah Mak, Thomas Chung Wai Mak, 2012-10-11 This book consists of over 300 problems and their solutions in structural inorganic chemistry at the senior undergraduate and beginning graduate level. The topics covered comprise Atomic and Molecular Electronic States, Atomic Orbitals, Hybrid Orbitals, Molecular Symmetry, Molecular Geometry and Bonding, Crystal Field Theory, Molecular Orbital Theory, Vibrational Spectroscopy and Crystal Structure. The central theme running through these topics is symmetry, molecular or crystalline. The problems collected in this volume originate in examination papers and take home assignments that have been part of the teaching of the book's two senior authors at The Chinese University of Hong Kong over the past four decades. The authors' courses include Chemical Bonding, Elementary Quantum Chemistry, Advanced Inorganic Chemistry, X-Ray Crystallography etc. The problems have been tested by generations of students taking these courses.

MATERIALS SCIENCE AND ENGINEERING : PROBLEMS WITH SOLUTIONS SHETTY, M.N., 2015-12-01 This book with analytical solutions to 260 select problems is primarily designed for the second year core course on materials science. The treatment of the book reflects the author's experience of teaching this course comprehensively at IIT Kanpur for a number of years to the students of engineering and 5 year integrated disciplines. The problems have been categorised into five sections covering a wide range of solid state properties. Section 1 deals with the dual representation of a wave and a particle and then comprehensively

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Introduction To Modern Physics: Solutions To Problems Paolo Amore, John Dirk Walecka, 2013-08-16 Our understanding of the physical world was revolutionized in the twentieth century the era of modern physics The book Introduction to Modern Physics Theoretical Foundations aimed at the very best students presents the foundations and frontiers of today's physics Typically students have to wade through several courses to see many of these topics The goal is to give them some idea of where they are going and how things fit together as they go along The book focuses on the following topics quantum mechanics applications in atomic nuclear particle and condensed matter physics special relativity relativistic quantum mechanics including the Dirac equation and Feynman diagrams quantum fields and general relativity The aim is to cover these topics in sufficient depth that things make sense to students and they achieve an elementary working knowledge of them The book assumes a one year calculus based freshman physics course along with a one year course in calculus Several appendices bring the reader up to speed on any additional required mathematics Many problems are included a great number of which take dedicated readers just as far as they want to go in modern physics The present book provides solutions to the over 175 problems in Introduction to Modern Physics Theoretical Foundations in what we believe to be a clear and concise fashion

Principles of Modern Chemistry David W. Oxtoby, H. Pat Gillis, Laurie J. Butler, 2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom

Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook McGraw Hill, 2001-08
Glencoe Chemistry Solving Problems A Chemistry Handbook Matter and Change

Dr. Salm's Chemistry Problem Drill Book Robert J. Salm, Robert J. Salm Robert J. Salm Ph.D., 2005 This drill book contains many common problem types that are asked in General Chemistry classes in High School and College This work will give you practice with the major problem types as you prepare for finals and standardized tests

Ebook: Chemistry: The Molecular Nature of Matter and Change
Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change

Ebook: Chemistry Julia Burdge, 2014-10-16 Chemistry Third Edition by Julia Burdge offers a clear writing style written with the students in mind Julia uses her background of teaching hundreds of general chemistry students per year and creates content to offer more

detailed explanation on areas where she knows they have problems With outstanding art a consistent problem solving approach interesting applications woven throughout the chapters and a wide range of end of chapter problems this is a great third edition text

Chemistry Problem Solver A. Lamont Tyler, Editors of REA, 2012-04-27 Each Problem Solver is an insightful and essential study and solution guide chock full of clear concise problem solving gems All your questions can be found in one convenient source from one of the most trusted names in reference solution guides More useful more practical and more informative these study aids are the best review books and textbook companions available Nothing remotely as comprehensive or as helpful exists in their subject anywhere Perfect for undergraduate and graduate studies Here in this highly useful reference is the finest overview of chemistry currently available with hundreds of chemistry problems that cover everything from atomic theory and quantum chemistry to electrochemistry and nuclear chemistry Each problem is clearly solved with step by step detailed solutions

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