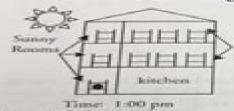
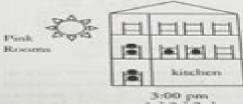
Learning Target - I understand that, when an atom is in ground state, that its electrons occupy the lowestenergy levels, sublevels, and orbitals.

Success Criteria - I can use the periodic table to determine the electron configuration for ground-state atoms.

Model 1 - The Boarding House



Manager's Code:



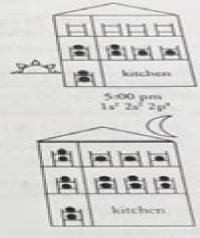
3:00 pm 1s² 2s² 2p³



Time: 7:00 pm Manager's Code: 1s 2s 2p 3s'



9:00 pm 1st 2st 2pt 3st 3pt



11:00 pm 1s' 2s' 2p' 3s' 3p'

1. Examine the boarding house diagrams in Model 1. Match each symbol below with its most likely meaning.

I. Bunk bed for boarders

II. Manager's code for the number of boarders in the house and their room assignments.

c. 1s2 2s2 2p2 3s2

III. Boarder

Dale Rich

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polyatomic species The following sequences are given C H CH H CH2 H CH3 H CH4 saturated and tetrahedral arrangement and N H NH H NH2 H NH3 H NH4 also saturated and tetrahedral arrangement The emphasis is to show trend from atomic structure through sp sp2 and sp3 hybridization Chapter 3 is composed of selections from Atomic and Molecular Configuration 1987 Included is MRAE AUFBAU TEMPLATE of the periodic table elements expanded to two pages It is informative to compare the MRAE concept for atoms to the MRAE method for diatomic species in Chapter one In both cases the key is the changes in electronic structure when an electron is added Electron Configuration, **Structure and Bonding** Primrose Kitten, 2017-06-19 I m constantly telling you the best way to learn is by practicing questions so I ve made you a book full of practice questions Multiple choice questions to reflect the style of exam questions activities to complete equations for you to balance compounds for you to work out the formula for lots of things that you need to recall and practice long answer exam style questions. This book is not designed as a text book or revision guide but as a workbook There are lots of good and bad expensive and free revision guides out there on my YouTube channel and other great websites So there is no point in me adding to the masses All the teaching all the new content is available for free on my YouTube channel this book is for you to practice and learn The best way to approach this is to watch the teaching video and make notes or after class try a section and check the answers Any corrections that are needed after the book is published will be listed on my website www primrosekitten com these will be corrected in the next version of the bookAnswers are provided for the sections where you need to work out the answers for yourself not the sections where you are just filling things in from a video or website Atoms Electrons Structure and Bonding Workbook Topics Covered are Some of this content has also been published in the Summer Start to A Level Chemistry and a practice exam paperAtomic Structure 20 Multiple choice questions Properties of Ionic Compounds 15 multiple choice questions Reference table of common ions formulae Formula of Ionic Compounds 65 formulas to work out Drawing Ionic Bonding 10 CompoundsSimple Covalent Bonding 20 multiple choice questionsDrawing Covalent Bonding 10 Compounds Summary Table for the 4 Different Types of Bonding Electron configurations Drawing electron configurations Drawing Electron Configurations Spot the mistake Electronic Configuration 20 multiple choice questions Exceptions to the Octet Rule Oxidation Numbers 20 multiple choice questions Balancing Equations using the oxidation numbers method 20 to practice Salt Equations 20 equations to complete and balance Shapes of Molecules Investigation Shapes of Molecules and Bond Angles 20 multiple choice questions Electronegativity and Bond Polarity Investigation Intermolecular Bonding 10 multiple choice questions Electrons Structure and Bonding Practice Exam **Electron Configuration as Ouestions Answers** *Electron Configuration in the Elements ...* Julius Sumner Miller,1933 the Basis of the Periodic Table William Fay Luder, 1943

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