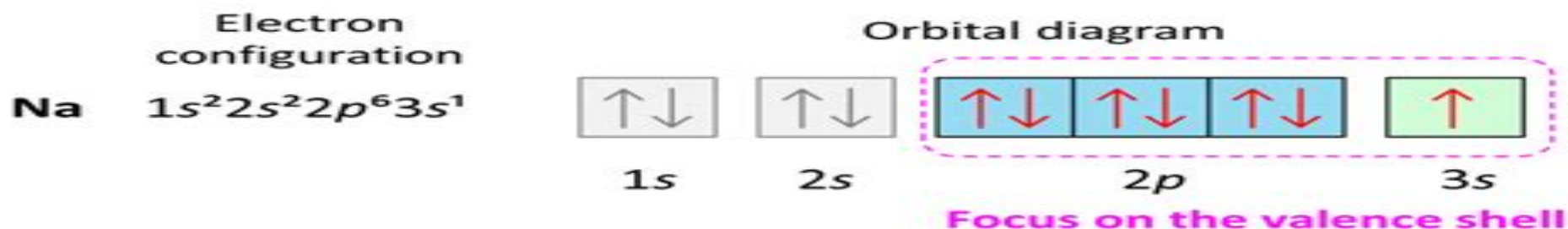


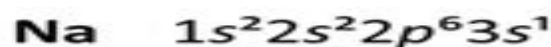
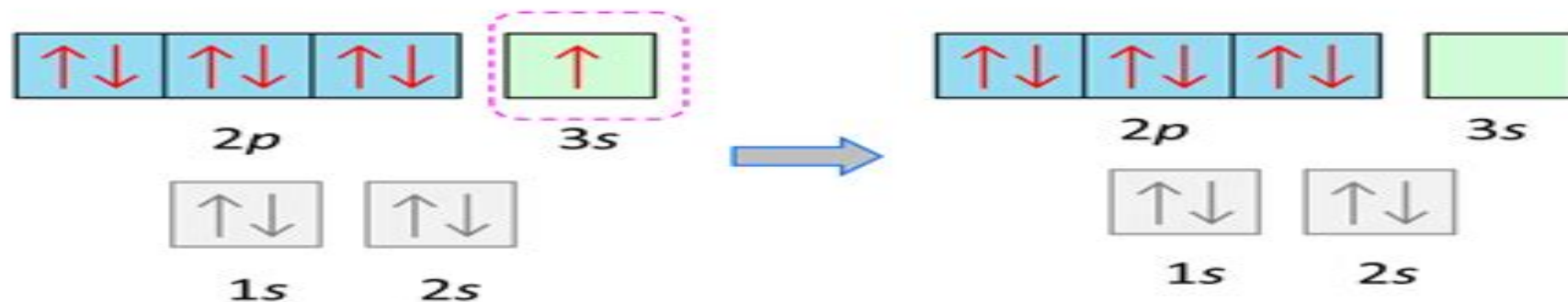
# Electron Configuration of Main Group Cations



The charge is +1, therefore, one electron is lost:



The electron must be from the outermost valence orbital –  $3s^1$



↓ Same as



Cations of main group metals attain the electron configuration of the noble gas that precedes them in the periodic table. They are **isoelectronic** with the previous noble gas.

# Electron Configuration Of Atoms And Ions Lab

**D Kirk**

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red circular gradient shape partially visible behind it.

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