

FLUID BALANCE

Fluid balance is a measure of the difference between fluid intake and fluid output.

Fluid Intake

- Water
- Food
- Medication
- Intravenous fluids
- Transfusion
- Other fluids

Fluid Output

- Urine
- Stool
- Sweat
- Respiratory
- Other

Fluid Balance

- Positive fluid balance: fluid intake is greater than fluid output.
- Negative fluid balance: fluid output is greater than fluid intake.
- Fluid balance is a key indicator of a patient's fluid status.

IV solutions

0.9% NaCl

- Isotonic
- Used for fluid resuscitation
- Used for fluid replacement

0.45% NaCl

- Hypotonic
- Used for fluid replacement
- Used for fluid resuscitation

5% D5W

- Hypotonic
- Used for fluid replacement
- Used for fluid resuscitation

0.9% NaCl + 5% D5W

- Isotonic
- Used for fluid replacement
- Used for fluid resuscitation

Intravenous Therapy Complications

Infusion

- Rate of infusion
- Volume of infusion
- Concentration of infusion

Infusion rate

- Too fast
- Too slow

Infusion volume

- Too much
- Too little

Infusion concentration

- Too high
- Too low

STOP

Electrolyte: Sodium

Sodium

Na⁺

135-145 mEq/L

Hypernatremia

- Increased level of sodium
- Causes: dehydration, excessive sodium intake

Hyponatremia

- Decreased level of sodium
- Causes: excessive water intake, renal failure

Interventions

- Hypernatremia:
 - Water deficit -> oral or IV water
 - Sodium deficit -> IV NaCl solution
 - Monitor sodium levels
- Hyponatremia:
 - Water excess -> fluid restriction
 - Sodium deficit -> IV NaCl solution
 - Monitor sodium levels

Electrolyte: Calcium

Calcium

Ca²⁺

9-10 mEq/L

Hypercalcemia

- Increased level of calcium
- Causes: hyperparathyroidism, multiple myeloma

Hypocalcemia

- Decreased level of calcium
- Causes: hypoparathyroidism, renal failure

Interventions

- Hypercalcemia:
 - With low sodium level, increase fluid intake to 2-3 L/day
 - Monitor sodium levels
- Hypocalcemia:
 - With low sodium level, increase fluid intake to 2-3 L/day
 - Monitor sodium levels

BALANCE

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Acid-Base Imbalances

Metabolic Acidosis

- Low pH
- Low HCO₃⁻
- Causes: lactic acidosis, renal failure, ketoacidosis

Metabolic Alkalosis

- High pH
- High HCO₃⁻
- Causes: vomiting, diuretic use

Respiratory Acidosis

- Low pH
- High PCO₂
- Causes: hypoventilation, COPD

Respiratory Alkalosis

- High pH
- Low PCO₂
- Causes: hyperventilation, anxiety

ARTERIAL BLOOD GAS (ABG) ANALYSIS

ABG Analysis

ABG analysis is a test that measures the levels of oxygen, carbon dioxide, and pH in the blood.

Normal ABG Values

- pH: 7.35-7.45
- PCO₂: 35-45 mmHg
- HCO₃⁻: 22-28 mEq/L

ABG Interpretation

- Acidosis: pH < 7.35
- Alkalosis: pH > 7.45
- Hypercapnia: PCO₂ > 45 mmHg
- Hypocapnia: PCO₂ < 35 mmHg
- Hypernatremia: Na⁺ > 145 mEq/L
- Hyponatremia: Na⁺ < 135 mEq/L

Root Cause

Root cause is the underlying condition that is causing the ABG abnormality.

Electrolyte: Magnesium

Magnesium

Mg²⁺

1.3-2.1 mEq/L

Hypermagnesemia

- Increased level of magnesium
- Causes: renal failure, excessive magnesium intake

Hypomagnesemia

- Decreased level of magnesium
- Causes: renal failure, excessive magnesium loss

Interventions

- Hypermagnesemia:
 - Stop magnesium intake
 - Monitor magnesium levels
- Hypomagnesemia:
 - Stop magnesium intake
 - Monitor magnesium levels

Electrolyte: Potassium

Potassium

K⁺

3.5-5 mEq/L

Hyperkalemia

- Increased level of potassium
- Causes: renal failure, excessive potassium intake

Hypokalemia

- Decreased level of potassium
- Causes: renal failure, excessive potassium loss

Interventions

- Hyperkalemia:
 - Stop potassium intake
 - Monitor potassium levels
- Hypokalemia:
 - Stop potassium intake
 - Monitor potassium levels

Electrolytes Cheat Sheet Acid Base Balance

D Kirk



Electrolytes Cheat Sheet Acid Base Balance:

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