
GCSE Chemistry – Electrolysis Exam Questions and Answers

What is electrolysis? - Answer- passing an electric current through an ionic substance that's molten or in a solution and breaks it down into new substances

What is electrolysis used for? - Answer- To make new substances

What is an electrolyte? - Answer- a liquid that conducts electricity

How are electrolytes made? - Answer- by melting or dissolving ionic compounds

Which part of an electrolyte conducts electricity? - Answer- The free electrons

What is the a positive electrode called? - Answer- Anode

What is a negative electrode called? - Answer- Cathode

What are electrodes made from and why? - Answer- Unreactive metal so they don't take part in the reaction

How do ions become atoms or molecules? - Answer- By losing or gaining electrons

How does an electrolysis circuit become complete? - Answer- A flow of electrons taken from anodes to cathodes.

How do you measure the conductivity of an electrolyte? - Answer- With a conductivity probe

In molten ionic compounds there is only _____ - Answer- one source of ions


Why can molten ionic compounds be electrolysed? - Answer- Because ions can move freely

In the electrolysis of molten lead bromide what is the half equation for the Pb^{2+} ions? - Answer- $\text{Pb}^{2+} + 2\text{e}^{-} \rightarrow \text{Pb}$

In the electrolysis of molten lead bromide what is the half equation for the Br^{-} ions? - Answer- $2\text{Br}^{-} \rightarrow \text{Br}_2 + 2\text{e}^{-}$

Electrolysis Questions And Answers

**S. Sivanagaraju, M. Balasubba
Reddy, D. Srilatha**



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Veterinary State Board Questions and Answers Victor Gage Kimball, 1917 **The Electrical Engineer**, 1898

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Teaching and Learning in the School Chemistry Laboratory Avi Hofstein, Muhamad Hugerat, 2021-11-05

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Commonly Asked Questions in Thermodynamics Marc J. Assael, Geoffrey C. Maitland, Thomas Maskow, Urs von Stockar, William A. Wakeham, Stefan Will, 2022-08-05 CRC Press is pleased to introduce the new edition of *Commonly Asked Questions in Thermodynamics* an indispensable resource for those in modern science and engineering disciplines from molecular science engineering and biotechnology to astrophysics Fully updated throughout this edition features two new chapters focused on energy utilization and biological systems This edition begins by setting out the fundamentals of thermodynamics including its basic laws and overarching principles It provides explanations of those principles in an organized manner using questions that arise frequently from undergraduates in the classroom as the stimulus These early chapters explore the language of thermodynamics the first and second laws statistical mechanical theory measurement of thermodynamic quantities and their relationships phase behavior in single and multicomponent systems electrochemistry and chemical and biochemical reaction equilibria The later chapters explore applications of these fundamentals to a diverse set of subjects including power generation with and without fossil fuels for transport industrial and domestic use heating decarbonization technologies energy storage refrigeration environmental pollution and biotechnology Data sources for the properties needed to complete thermodynamic evaluations of many processes are included The text is designed for readers to dip into to find an answer to a specific question where thermodynamics can provide some if not all of the answers whether in the context of an undergraduate course or not Thus its readership extends beyond conventional technical undergraduates to practicing engineers and also to the interested lay person who seeks to understand the discourse that surrounds the choice of particular technological solutions to current and future energy and material production problems **Hygeia** , 1927

Chemistry-vol-I Dr Vijay Sarda, Dr A C Handa, Dr K K Arora, A text book on Chemistry *Answers to questions set at the science and art examinations* Science and art department, 1887

Reviewing **Electrolysis Questions And Answers**: Unlocking the Spellbinding Force of Linguistics

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