



Supported by:

Pencil Electrolysis

Example Indiana Science Indicators Addressed:

- 1.1.1 Observe, describe, draw and sort objects carefully to learn about them.
- 1.1.2 Investigate and make observations to seek answers to questions about the world.
- 3.1.4 Discuss the results of investigations and consider the explanations of others.
- 4.2.3 Make simple and safe electrical connections with various sockets, plugs and terminals.
- 6.3.19 Investigate that materials may be composed of parts that are too small to be seen without magnification.

Materials:

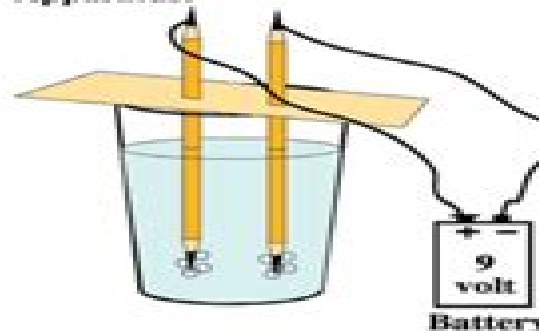
- Two regular number 2 pencils (remove eraser and metal part on the ends)
- 2 transparent plastic cups per group
- water
- table salt (NaCl)
- spoon (for mixing)
- 1 9V battery/group
- 2 pieces of cardboard (enough to cover the opening)
- 2 connecting wires (optional)
- tape (optional)

Caution: Use only 9V batteries (direct current) for this experiment.

Procedure:

- Sharpen each pencil at both ends.
- Cut the cardboard to fit over glass.
- Push the two pencils into the cardboard, about 1.5 centimeters apart.
- Dissolve about a teaspoon of salt into the warm water and let sit for a while. The salt helps conduct the electricity better in the water.
- (Option 1) Using one piece of the electrical wire, connect one end on the positive side of the battery and the other to the black graphite (the "lead" of the pencil) at

Apparatus:



http://www.energyquest.ca.gov/projects/split_h2o.html

Electrochemistry Pencil Electrolysis Lab Answers

Y Pai



Electrochemistry Pencil Electrolysis Lab Answers:

This book delves into Electrochemistry Pencil Electrolysis Lab Answers. Electrochemistry Pencil Electrolysis Lab Answers is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Electrochemistry Pencil Electrolysis Lab Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Electrochemistry Pencil Electrolysis Lab Answers
- Chapter 2: Essential Elements of Electrochemistry Pencil Electrolysis Lab Answers
- Chapter 3: Electrochemistry Pencil Electrolysis Lab Answers in Everyday Life
- Chapter 4: Electrochemistry Pencil Electrolysis Lab Answers in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Electrochemistry Pencil Electrolysis Lab Answers. This chapter will explore what Electrochemistry Pencil Electrolysis Lab Answers is, why Electrochemistry Pencil Electrolysis Lab Answers is vital, and how to effectively learn about Electrochemistry Pencil Electrolysis Lab Answers.
3. In chapter 2, this book will delve into the foundational concepts of Electrochemistry Pencil Electrolysis Lab Answers. The second chapter will elucidate the essential principles that must be understood to grasp Electrochemistry Pencil Electrolysis Lab Answers in its entirety.
4. In chapter 3, this book will examine the practical applications of Electrochemistry Pencil Electrolysis Lab Answers in daily life. The third chapter will showcase real-world examples of how Electrochemistry Pencil Electrolysis Lab Answers can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Electrochemistry Pencil Electrolysis Lab Answers in specific contexts. This chapter will explore how Electrochemistry Pencil Electrolysis Lab Answers is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Electrochemistry Pencil Electrolysis Lab Answers. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Electrochemistry Pencil Electrolysis Lab Answers.

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