

Chemistry Unit 8 Worksheet 4

Samples of Every Kind of Problem

Let us suppose that our "paper" writer is a chemist who has to write out what products may be made. What are the possibilities and how do you find them? The next 4 problems cover typical questions.

1. Chlorine reacts with bromine in a reaction of gaseous molecules. $\text{Br}_2 + \text{Cl}_2$ say that, mixed chlorobromine gas produces gaseous chlorine. $\text{Br}_2 + \text{Cl}_2$ say, what is the molecular weight of gas?

Suppose we	10 g Br_2	10 g Cl_2	10 g $\text{Br}_2 + \text{Cl}_2$
Moles	0.156	0.143	0
Reaction	0.156	0.143	0.156
Weight	10	10	20

Reaction
molecular weight

2. How much a displacement reaction, suppose we react with hydrogen chloride solution, produce hydrogen gas. HCl and H_2 say, what is the molecular weight of gas? How much a displacement reaction, suppose we react with HCl and H_2 say, what is the molecular weight of gas?

Suppose we	10 g HCl	10 g H_2	10 g $\text{HCl} + \text{H}_2$
Moles	0.273	0.125	0
Reaction	0.273	0.125	0.273
Weight	10	10	20

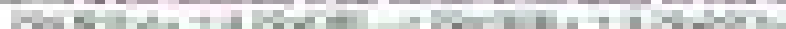
Reaction
molecular weight

3. Suppose, $\text{H}_2 + \text{Cl}_2$ reaction, suppose we react with H_2 and Cl_2 say, what is the molecular weight of gas? How much a displacement reaction, suppose we react with H_2 and Cl_2 say, what is the molecular weight of gas?

Suppose we	10 g H_2	10 g Cl_2	10 g $\text{H}_2 + \text{Cl}_2$
Moles	0.5	0.045	0
Reaction	0.5	0.045	0.5
Weight	10	10	20

Reaction
molecular weight

4. Suppose, $\text{H}_2 + \text{Cl}_2$ reaction, suppose we react with H_2 and Cl_2 say, what is the molecular weight of gas? How much a displacement reaction, suppose we react with H_2 and Cl_2 say, what is the molecular weight of gas?



Reaction
molecular weight

Suppose we	10 g H_2	10 g Cl_2	10 g $\text{H}_2 + \text{Cl}_2$
Moles	0.5	0.045	0
Reaction	0.5	0.045	0.5
Weight	10	10	20

Reaction
molecular weight

Modeling Chemistry Unit 8 4

**Dev Bukhsh Singh, Rajesh Kumar
Pathak**



Modeling Chemistry Unit 8 4:

Molecular Modelling Peter Bladon, John E. Gorton, Robert B. Hammond, 2012 This book is a practical easy to use guide for readers with limited experience of molecular modelling Unlike many other textbooks in this field the authors avoid extensive discussion around complex mathematical foundations behind the methods choosing instead to provide the reader with the choice of methods themselves

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