

Chemistry Unit 8 Worksheet 4

Samples of Every Kind of Problem

Let us suppose that during "paper", writing a composition about how the world will change people may disagree. However, this particular one based on physics. The main 3 numbers given (based on mass):

1. Chlorine gas (the substance) is heated to produce hydrogen gas. HClO_2 and H_2 are mixed substances for producing hydrogen gas. HClO_2 is used. 1.0 mol of hydrogen gas is given.

Substance	HClO_2	H_2	HClO_2	H_2
Mass	1.0 g	1.0 g	1.0 g	1.0 g
Volume	1.0 L	1.0 L	1.0 L	1.0 L
Temperature	100°C	100°C	100°C	100°C

1.0 mol of HClO_2 is given

2. Chlorine gas (the substance) is heated to produce hydrogen gas. HClO_2 and H_2 are mixed substances for producing hydrogen gas. HClO_2 is used. 1.0 mol of hydrogen gas is given.

Substance	HClO_2	H_2	HClO_2	H_2
Mass	1.0 g	1.0 g	1.0 g	1.0 g
Volume	1.0 L	1.0 L	1.0 L	1.0 L
Temperature	100°C	100°C	100°C	100°C

1.0 mol of HClO_2 is given

3. Chlorine gas (the substance) is heated to produce hydrogen gas. HClO_2 and H_2 are mixed substances for producing hydrogen gas. HClO_2 is used. 1.0 mol of hydrogen gas is given.

Substance	HClO_2	H_2	HClO_2	H_2
Mass	1.0 g	1.0 g	1.0 g	1.0 g
Volume	1.0 L	1.0 L	1.0 L	1.0 L
Temperature	100°C	100°C	100°C	100°C

1.0 mol of HClO_2 is given

4. Chlorine gas (the substance) is heated to produce hydrogen gas. HClO_2 and H_2 are mixed substances for producing hydrogen gas. HClO_2 is used. 1.0 mol of hydrogen gas is given.

HClO_2 is used. 1.0 mol of hydrogen gas is given.

Substance	HClO_2	H_2	HClO_2	H_2
Mass	1.0 g	1.0 g	1.0 g	1.0 g
Volume	1.0 L	1.0 L	1.0 L	1.0 L
Temperature	100°C	100°C	100°C	100°C

1.0 mol of HClO_2 is given

Modeling Chemistry Unit 8 2 Answers

Yicheng Fang



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