

1.14b How to Count Atoms

1. e.g. Na = 1 atom of sodium
2. e.g. H_2 = 2 atoms of hydrogen
3. e.g. $\text{Mg}_3(\text{PO}_4)_2$ = 1 molecule of magnesium phosphate
 3 atoms of magnesium
 2 atoms of phosphorus
 8 atoms of oxygen
 Total # of atoms = 13
4. a) 3 C = 3 atoms of carbon
 e.g. not 3C
 b) $2 \text{ H}_2\text{O}$ = 2 molecules of water
 4 atoms of hydrogen
 2 atoms of oxygen
 Total # of atoms = 6
 c) 3 CuSO_4 = 3 molecules of sulfate
 3 atoms of copper
 3 atoms of sulfur
 12 atoms of oxygen
 Total # of atoms = 18
 d) $4 \text{ Pb}(\text{NO}_3)_2$ = 4 molecules of lead nitrate
 4 atoms of lead
 8 atoms of nitrogen
 24 atoms of oxygen
 Totals # of atoms = 36

1.14c Counting Atoms Worksheet



Type of Atom	# of Atoms
sodium	2
carbon	1
oxygen	3
Total	6



Type of Atom	# of Atoms
calcium	3
phosphorus	2
oxygen	8
Total	13



Type of Atom	# of Atoms
potassium	2
chromium	1
oxygen	4
Total	7



Type of Atom	# of Atoms
barium	3
chlorine	6
Total	9



Type of Atom	# of Atoms
nitrogen	1
hydrogen	7
carbon	2
oxygen	2
Total	12



Type of Atom	# of Atoms
aluminum	8
carbon	12
oxygen	36
Total	56



Type of Atom	# of Atoms
lead	1
nitrogen	2
oxygen	6
Total	9



Type of Atom	# of Atoms
nitrogen	4
hydrogen	16
chromium	4
oxygen	14
Total	38



Modeling Chemistry Worksheet Rearranging Atoms

Answer Key

Wenbin Ji



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