

Chemistry – Unit 11 – Worksheet 2

Periodic trends

1. On the graph of the first ionization energy (IE₁) vs. the atomic number (Z) for the first 20 elements, label each point with the symbol of the element it represents.
2. On the graph of the atomic radius (in nm) vs. the atomic number, label each point with the symbol of the element it represents.

Group = _____

Period = _____

Conclusion

1. Examine the graph of 1st ionization energy vs. atomic number. Can a periodic trend be observed? If so, describe the trend.
2. Which group (or column #) has the highest ionization energies? Which group has the lowest?
3. Examine the graph of atomic radius vs. atomic number. Can a periodic trend be observed? If so, describe the trend.
4. For a given period (a row in the Periodic Table), which group (a column in the table) appears to have the largest atomic radii? Which group has the smallest?
5. What appears to be the relationship between trends in IE and atomic radii?

Modeling Chemistry Unit 11 Ws 2 V1

S Marginson



Modeling Chemistry Unit 11 Ws 2 V1:

The book delves into Modeling Chemistry Unit 11 Ws 2 V1. Modeling Chemistry Unit 11 Ws 2 V1 is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Modeling Chemistry Unit 11 Ws 2 V1, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Modeling Chemistry Unit 11 Ws 2 V1
- Chapter 2: Essential Elements of Modeling Chemistry Unit 11 Ws 2 V1
- Chapter 3: Modeling Chemistry Unit 11 Ws 2 V1 in Everyday Life
- Chapter 4: Modeling Chemistry Unit 11 Ws 2 V1 in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Modeling Chemistry Unit 11 Ws 2 V1. The first chapter will explore what Modeling Chemistry Unit 11 Ws 2 V1 is, why Modeling Chemistry Unit 11 Ws 2 V1 is vital, and how to effectively learn about Modeling Chemistry Unit 11 Ws 2 V1.

3. In chapter 2, this book will delve into the foundational concepts of Modeling Chemistry Unit 11 Ws 2 V1. This chapter will elucidate the essential principles that must be understood to grasp Modeling Chemistry Unit 11 Ws 2 V1 in its entirety.

4. In chapter 3, this book will examine the practical applications of Modeling Chemistry Unit 11 Ws 2 V1 in daily life. This chapter will showcase real-world examples of how Modeling Chemistry Unit 11 Ws 2 V1 can be effectively utilized in everyday scenarios.

5. In chapter 4, the author will scrutinize the relevance of Modeling Chemistry Unit 11 Ws 2 V1 in specific contexts. The fourth chapter will explore how Modeling Chemistry Unit 11 Ws 2 V1 is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about Modeling Chemistry Unit 11 Ws 2 V1. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Modeling Chemistry Unit 11 Ws 2 V1.

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