# Excerpt from Activity 3B Exploration Questions

# Model 1: 1st Definition of Redox Reactions

Reduction Reactions

$$2NO_2 \longrightarrow N_2 + 2O_2$$

$$SO_2 \longrightarrow S + O_2$$

$$2CO \longrightarrow 2C + O_2$$

- Under the reduction reactions of Model 1, what are the products of:
  - A) the reduction of CO?
    2C and O<sub>2</sub>
  - B) the reduction of SO<sub>2</sub>?
    S and O<sub>2</sub>
  - C) the reduction of 2NO<sub>2</sub>?
    N<sub>2</sub> and 2O<sub>2</sub>

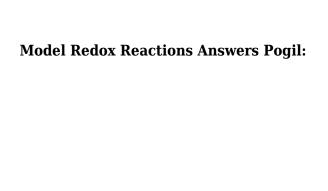
Oxidation Reactions

$$N_2 + 2O_2 \rightarrow 2NO_2$$
  
 $S + O_2 \rightarrow SO_2$   
 $2C + O_2 \rightarrow 2CO$ 

2. Using your answers to question 1, what do the reduction reactions of Model 1 have in common? An O<sub>2</sub> molecule is produced in each reduction reaction.

# **Model Redox Reactions Answers Pogil**

Joacim Rocklöv



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