



Chemistry Chapter 10

Name _____

Stoich Worksheet 2

Stoichiometry Practice Problems 1

Do this on a separate sheet of paper. **Show all work and use units.** Round your answers to 3 significant digits.

1. What mass of iron is needed to react completely with 64.2 g of sulfur?



2. How much copper (I) oxide is needed to produce 75.0 g of copper (I) chloride?



3. How much hydrogen sulfite is produced when 178 g of sulfur dioxide reacts with water?



4. If 65.0 g of iron (II) sulfide is added to hydrogen chloride, what mass of hydrogen sulfide is produced?



5. Silver bromide is a precipitate that is formed when silver nitrate reacts with sodium bromide. What mass of the precipitate is produced if you start with 48.7 g of silver nitrate?



6. How much hydrogen nitrate is needed to react completely with 36.3 g of magnesium?



7. What mass of hydrogen can be produced when 82.1 g of zinc reacts with hydrogen sulfate?



For #8 & 9, you will need to use the density of oxygen gas is 1.43g/L.

8. What volume of oxygen is produced when 28.9g of water is decomposed?



9. What volume of oxygen is produced from the decomposition of 145.7g of potassium chlorate?



Mixed Stoichiometry Word Problems

RJ Alexander



Mixed Stoichiometry Word Problems:

Unveiling the Energy of Verbal Artistry: An Psychological Sojourn through **Mixed Stoichiometry Word Problems**

In some sort of inundated with displays and the cacophony of fast conversation, the profound power and psychological resonance of verbal art often disappear into obscurity, eclipsed by the regular barrage of noise and distractions. However, located within the lyrical pages of **Mixed Stoichiometry Word Problems**, a fascinating function of fictional brilliance that impulses with natural emotions, lies an remarkable journey waiting to be embarked upon. Penned by way of a virtuoso wordsmith, that magical opus books visitors on a psychological odyssey, lightly revealing the latent potential and profound affect stuck within the complex internet of language. Within the heart-wrenching expanse of this evocative evaluation, we will embark upon an introspective exploration of the book is main subjects, dissect their interesting publishing design, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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