

Nomenclature Worksheet 3 – Covalent (Molecular) Compounds

Part A: Name the following covalent compounds.

1. CO _____
2. CO_2 _____
3. N_2O_5 _____
4. N_2 _____
5. NP _____
6. SCl_2 _____
7. P_2O_5 _____
8. NBr_3 _____
9. Cl_4 _____
10. OCl_4 _____
11. PF_3 _____
12. PF_5 _____
13. OS _____
14. SeP_2 _____
15. TeBr_2 _____
16. P_2S_5 _____
17. C_3N_2 _____
18. F_2 _____
19. CH_4 _____
20. PH_3 _____

Part B: Write the Chemical Formula for each of the following compounds.

1. carbon tetrafluoride _____
2. silicon dioxide _____
3. dinitrogen trisulfide _____
4. phosphorus mononitride _____
5. hydrogen gas _____
6. carbon disulfide _____
7. nitrogen trichloride _____
8. silicon tetrabromide _____
9. carbon dioxide _____
10. nitrogen trichloride _____
11. boron trisulfide _____
12. sulphur trioxide _____
13. selenium tetrafluoride _____
14. diphosphorus pentasulfide _____
15. xenon tetrafluoride _____
16. sulfur dibromide _____
17. carbon tetrachloride _____
18. oxygen gas _____
19. fluorine gas _____
20. dinitrogen tetroxide _____

Mixed Ionic Compound Homework

Siddappa N.Byrareddy



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