

Question Number	Acceptable Answers	Reject	Mark
23 (d)(i)	$2\text{SO}_2(\text{g}) + 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{SO}_4(\text{aq})$ <p>ALLOW Multiples</p> <p>M1 - Species and balancing (1)</p> <p>M2 - All state symbols correct</p> <p>M2 can only be awarded for the correct state symbols if M1 has already been awarded OR for a 'near-miss' equation with the species almost correct</p> <p>(1)</p>		2

Question Number	Acceptable Answers	Reject	Mark
23 (d)(ii)	<p>CaO is basic / is a base / is a metal oxide OR CaO neutralizes (acidic) SO_2 / H_2SO_4 OR CaO reacts with a non-metal oxide (SO_2) OR Basic oxides react with acidic gases</p> <p>ALLOW Alkaline for basic/ alkali for base</p> <p>IGNORE References to forming a salt / formation of calcium sulfate</p> <p>References to the large surface area of powder / effect on rate of reaction</p>		1

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TD Snyder



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