

HP EliteBook 845 G8 Notebook PC (43M05UT)



Easily manage your busy day

Confidently connect and stay productive in almost any setting with a beautifully designed and highly secure HP EliteBook 845. Designed for the multi-task, multi-place workday, this thin and light PC is built for how you work.



HP recommends Windows 10 Pro for business

Designed for how you work

The HP EliteBook 845 is built for how you work today. This easy-to-carry ultrathin and thin business PC comes with an 85-percent screen-to-body ratio and a quiet and comfortable keyboard to work on.

Work with peace of mind

Security features from HP work together to create an always-on, always-acting, resilient defense. From the BIOS to the browser, above and below the OS, these constantly evolving solutions help protect your PC from threats.

Power, connectivity, long battery life

How we work is evolving quickly. Help drive connectivity and productivity with a premium AMD Ryzen™ Processor^[2], gigabit-speed Wi-Fi 6^[4,5], and long battery life.

Featuring

Let nothing stand in your way

Power through your day with Windows 10 Pro and the powerful security, collaboration and connectivity features from HP.^[1]

HP Sure Sense

Malware is evolving rapidly and traditional antivirus can't always recognize new attacks. Protect your PC against never-before-seen attacks with HP Sure Sense, which uses deep learning AI to provide exceptional protection against advanced malware.

Fast and efficient wireless LAN

The portability of your PC and the reliability of a fast connection determines where you can work. Get a fast and reliable connection in dense wireless environments with gigabit-speed Wi-Fi 6.^[5]

Automatic recovery from firmware attacks

Firmware attacks can completely devastate your PC. Stay protected with HP's self-healing BIOS – HP Sure Start Gen6 automatically recovers the BIOS from malware, rootkits, or corruption.

Browse confidently

Help protect your PC from websites and read-only Microsoft Office and PDF attachments with embedded malware, ransomware, or viruses with hardware-enforced security from HP Sure Click.^[6]

Your personal coach

Make your business more profitable and get on track faster at work with HP WorkWatt. Designed for coaches and

Hp Pavilion Dv10notebook User Manual

Patrick Vollmar



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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid H_2CO_3 and its conjugate base the bicarbonate ion HCO_3^-

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to soap in this chapter we will examine the properties of acids and bases and learn about the chemical nature of these important compounds we will cover ph and how to calculate the ph of a solution

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acids bases and enzymes many acids and bases in living things provide the ph that enzymes need enzymes are biological catalysts that must work effectively for biochemical reactions to occur most enzymes can do their job only at a certain level of acidity cells secrete acids and bases to maintain the proper ph for enzymes to do their work

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acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen ion from an acid acidic substances are usually identified by their sour taste

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a binary acid is an acid that consists of hydrogen and one other element the most common binary acids contain a halogen the acid name begins with the prefix hydro followed by the base name of the anion followed by the suffix ic

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oct 18 2023 most of the acids are solid it has a sour taste they produce hydrogen after reacting with active metals like zinc magnesium aluminium or iron acids have a low ph value i e below 7 salts and water are produced when an acid reacts with base or alkalis most of the acids are solids

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arrhenius s definition of acids and bases the earliest definition of acids and bases is arrhenius s definition which states that an acid is a substance that forms hydrogen ions h when dissolved in water and a base is a substance that forms hydroxide

ions oh when dissolved in water

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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other s chemistry to produce a rather innocuous substance water in fact the general reaction between an acid and a base is ce acid base water salt nonnumber

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