CHEMISTRY

Chemically strong acid is:

Acetic acid

(a)

(a)

(14)

2014

Max. Marks: 17 Time: 30 Minutes SECTION "A" (MULTIPLE CHOICE QUESTIONS) Choose the correct answer for each from the given options: The branch of Chemistry which deals with the emission (1) of radiations from nuclei is called: Organic Chemistry Inorganic Chemistry (b) (a) Physical Chemistry **Nuclear Chemistry** (d) (c) Molecular mass expressed in grams is called: (2)Molar mass (b) (a) ·Atomic mass Equivalent mass Formula mass (d) (c) Which one of the following elements has its electronic (3) configuration K²L⁸M²? Magnesium (Mg) (a) Calcium (Ca) (b) (d) Sodium (Na) Neon (Ne) (c) It has the highest value of electro negativity: (4)Flourine (b) Chlorine (c) Bromine (d) Iodine (a) This bond is also called electrovalent bond: (5) (b) lonic bond (a) Covalent bond (d) Hydrogen bond Polar bond (c) The state of matter, having definite shape and volume is (6) known as: Solid (b) Liquid (c) gas (d) Plasma (a) If 0.4 is the mole fraction of solute, the mole fraction of (7) solvent would be: (a) 0.1 (b) 0.6 (c) 0.9 (d) 1.0 Which one of the following is an electrolyte? (8) Petrol (b) Ethyl Alcohol (c) Benzene (d) Sodium Chloride (a) The pH value of distilled water is: (9) 14 (d) (b) 4 (c) (a) It is an exothermic reaction: (10)Decomposition (a) Combustion (b) Fusion Photosynthesis (d) (c) It is called Ordinary Hydrogen: (11)Nascent Hydrogen (b) Protium (a) (d) Deuterium (c) Kaolin (China Clay) is used to make: (12)Crockery (b) Glass (c) Ceramics (d) Asbestos (a) The air we breathe in, usually contains the highest (13)proportion of:

Hydrogen (b) Nitrogen (c) Oxygen (d) Carbon dioxide

(b)

Nitric acid

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