

CHAPTER 4 REVIEW*Arrangement of Electrons in Atoms***SECTION 1**

SHORT ANSWER Answer the following questions in the space provided.

1. In what way does the photoelectric effect support the particle theory of light?

In order for an electron to be ejected from a metal surface, the electron must be struck by a single photon with at least the minimum energy needed to knock the electron loose.

2. What is the difference between the ground state and the excited state of an atom?

The ground state is the lowest energy state of the atom. When the atom absorbs energy, it can move to a higher energy state, or excited state.

3. Under what circumstances can an atom emit a photon?

A photon is emitted when an atom moves from an excited state to its ground state or to a lower-energy excited state.

4. How can the energy levels of the atom be determined by measuring the light emitted from an atom?

When an atom loses energy, it falls from a higher energy state to a lower energy state. The frequency of the emitted light, observed in an element's line-emission spectrum, may be measured. The energy of each transition is calculated using the equation $E = h\nu$, where ν is the frequency of each of the lines in the element's line-emission spectrum. From the analysis of these results, the energy levels of an atom of each element may be determined.

5. Why does electromagnetic radiation in the ultraviolet region represent a larger energy transition than does radiation in the infrared region?

Energy is proportional to frequency, and ultraviolet radiation has a higher frequency than infrared radiation. To produce ultraviolet radiation, electrons must drop to lower energy levels than they do to produce infrared radiation.

Electrons In Atoms Vocabulary Review Answers

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developing the landmark series of workbooks and applications to teach this Insider's Language to students around the world

Chemistry (Student) Dennis Englin, 2018-02-08 What is chemistry It is the study of the composition structure and properties of matter It is through an understanding of chemistry that the products that have benefited society were discovered and technologies to sustain the environment were put in place Knowledge taught in this course of how matter changes will give us an insight into the origin of life so we can realize that life could only have been formed by a supernatural act of creation not by a process of change over time High school science course with lab curriculum Lab experiments are included with step by step images for guidance Based on the principle that those who can understand and apply information do much better than those who simply memorize material This course has been taught by Dr Englin for many years with students going on to medical and graduate school He wanted to develop a series of courses that would give students the tools to help them succeed in higher education The comprehensive material has God the Creator as its foundation A teacher guide is available for Chemistry providing this full year science course with a detailed schedule worksheets and tests *Merrill*

Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 Atoms, Molecules & Elements: The Periodic Table Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice The Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives **Physical Science Junior High School Science Series 1986**, 1986-06 Discovering Science Through Inquiry: Matter Kit Rachel E. Green, 2010-05-12 The Discovering Science through Inquiry series provides teachers and students of grades 3-8 with direction for hands on science exploration around particular science topics and focuses The series follows the 5E model engage explore explain elaborate evaluate The Matter kit provides a complete inquiry model for the exploration of the structure and properties of matter through supported investigation Encourage students through activities such as studying the chemical properties of matter and investigating whether household items are acids and bases Matter kit includes 16 Inquiry Cards in print and digital formats Teacher's Guide Inquiry Handbook Each kit includes a single copy additional copies can be ordered Digital resources include PDFs of activities and additional teacher resources including images and assessment tools leveled background pages for students and video clips to support both students and teachers *Organic Chemistry* David R. Klein, 2020-12-22 In Organic Chemistry 4th

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Science insights Michael A. Dispezio, Marylin Lisowski, Gerald Duane Skoog, Marylin Linner-Luebe, Bobbie Sparks, 1996

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