CHARTER A REVOEW

## Arrangement of Electrons in Atoms

#### SECTION 2

SHORT ANSWER Asswer the following questions in the space provided.

- How many quantum numbers are used to describe the properties of electrons in atomic orbitals?
  - 460 T. Red 13 **海通数三海 656年一堂**
- A subcricial electron cloud surrounding un atomic nucleus would best represent.
  - days can a deficital. (c) a combination of two different p orbitals. districted problems. (d) a combination of an s and a a orbital.
- How many electrons can an energy level of it = 4 hold?
  - 1000 80 400 - 302400 24 E **多数的工作**
- How many electrons can an energy level of n = 2 hold?
  - 440 307 子说话:一次 4964 (204) Sept. 16.
- Companied with an electron for which n = 2, an electron for which n = 4Busic colores
  - Gall appear. CONTRACTOR ... (b) particle nature. 445 wave nature.
- According to Bohr, which is the point in the figure below where ofactrons camoot reside?"
  - day) modest A York, Intelliget Co. 4994 Profest III 445 motion (I)



- According to the quantum theory, point D in the above figure represents:
  - (a) the fixed position of an electron.
  - 651 the furthest position from the nucleus that an electron can achieve.
  - (c) is position where an electron probably exists.
  - dd) a position where as electron cannot exist.

# **Electrons In Atoms Chapter Test A Answers**

**RP Manchanda** 

## **Electrons In Atoms Chapter Test A Answers:**

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