

Electron Dot Diagram Periodic Table

Use a periodic table to construct Electron Dot Diagrams (Lewis Structure) for elements with atomic numbers 1-18.

Each box is worth one point.

1 H • H							2 He •• He
3 Li • Li	4 Be •• Be	5 B •• • B	6 C •• •• C	7 N •• •• • N	8 O •• •• •• O	9 F •• •• •• • F	10 Ne •• •• •• •• Ne
11 Na • Na	12 Mg •• Mg	13 Al •• • Al	14 Si •• •• •• Si	15 P •• •• •• • P	16 S •• •• •• •• S	17 Cl •• •• •• •• • Cl	18 Ar •• •• •• •• Ar

Follow-up Questions:

1. List your observations about the periodic table above: (4 points)

Suggested Answers: There are patterns. Elements in the same column (group/family) have the same number of dots (valence electrons). Elements in Group 1 have one valence electron; elements in Group 2 have two valence electrons; elements in Group 13 have three valence electrons; elements in Group 14 have four valence electrons, etc. As you move across the period from left to right, one dot (electron in the last energy level) is added.

Electrons And The Periodic Table Answer Key

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