

1. Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:

Cu   K   Ni   Br

- a. Explain why you made these choices: All of the elements are in the same period. The trend in atomic radius as you go across a period is DECREASING. Therefore, the element on the far left (K) is the largest, and the element on the far right (Br) is the smallest.

2. Circle the element with the highest ionization energy and put a square around the element with the lowest ionization energy:

Cu   K   Ni   Br

- a. Explain why you made these choices: All of the elements are in the same period. The trend in ionization energy as you go across a period is INCREASING. Therefore, the element on the far left (K) has the lowest ionization energy, and the element on the far right (Br) has the highest ionization energy.

3. Circle the element with the highest electronegativity and put a square around the element with the lowest electronegativity:

Cu   K   Ni   Br

- a. Explain why you made these choices: All of the elements are in the same period. The trend in electronegativity as you go across a period is INCREASING. Therefore, the element on the far left (K) has the lowest electronegativity, and the element on the far right (Br) has the highest electronegativity.

4. For each of the following groups: Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:

- 5.
- a. O   C   Be   Ne   Same Period
- b. Na   Rb   Fr   H   Same Group
- c. Pb   C   Sn   Si   Same Group
- d. Au   W   S   Fr   Ne   Zn   Challenge

6. For each of the following groups: Circle the element with the highest ionization energy and put a square around the element with the lowest ionization energy:

- a. O   C   Be   Ne   Same Period
- b. Na   Rb   Fr   H   Same Group
- c. Pb   C   Sn   Si   Same Group
- d. Au   W   S   Fr   Ne   Zn   Challenge

# Electronegativity Periodic Trends Practice Answers

**M Lipman**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## **Electronegativity Periodic Trends Practice Answers:**

## Whispering the Techniques of Language: An Psychological Quest through **Electronegativity Periodic Trends Practice Answers**

In a digitally-driven earth where monitors reign supreme and quick transmission drowns out the subtleties of language, the profound techniques and psychological subtleties concealed within phrases often go unheard. Yet, set within the pages of **Electronegativity Periodic Trends Practice Answers** a charming fictional value blinking with fresh feelings, lies an exceptional quest waiting to be undertaken. Written by a skilled wordsmith, that enchanting opus attracts visitors on an introspective trip, lightly unraveling the veiled truths and profound impact resonating within the very fabric of each word. Within the psychological depths of the poignant review, we will embark upon a sincere exploration of the book is key subjects, dissect its captivating publishing design, and succumb to the effective resonance it evokes serious within the recesses of readers hearts.

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