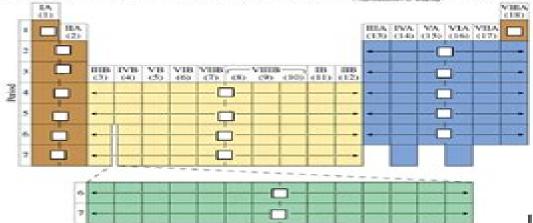
#### Objectives:

- Explain the nature of light and how it reveals the electron structure of the atom.
- Write the electron structure of elements
- Relate electron structure to the periodic table
- 1. Light is a form of what type of energy?
  - · Give 2 examples of other forms of this type of energy.
- 2. What is the meaning of wavelength and frequency of light?
- White light produces a continuous spectrum. Elements that are excited produce a line spectrum.
  What is a line spectrum?
- Different metals emit different colored light in flame tests. This light originates from inside the metal atoms. Explain what was happening inside these atoms that causes the various colors of light. Be specific
- 5. What is an orbital?
- List the orbitals that exist in the following energy levels. Indicate the number of these orbitals. n = 1, n = 2, n = 3, n = 4
- Indicate the electron configuration of these elements using the shorthand notation and the orbital diagrams.

For example - helium: 1s2, 1s

Elements: Li, C, Mg, Fe, Kr

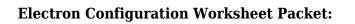
8. In this periodic chart, indicate the outer orbitals being filled with electrons in the rectangles.



9. Why do elements in the same group have similar properties?

# **Electron Configuration Worksheet Packet**

N.B. Singh



Unveiling the Magic of Words: A Report on "Electron Configuration Worksheet Packet"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Electron Configuration Worksheet Packet**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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