

Name: _____ Date: _____

Electron Arrangements with answer key

- What are electron shells?

Answer:

Electron shells are regions around the nucleus of an atom where electrons are likely to be found. Each shell can hold a specific number of electrons.

- How many electrons can each shell hold?

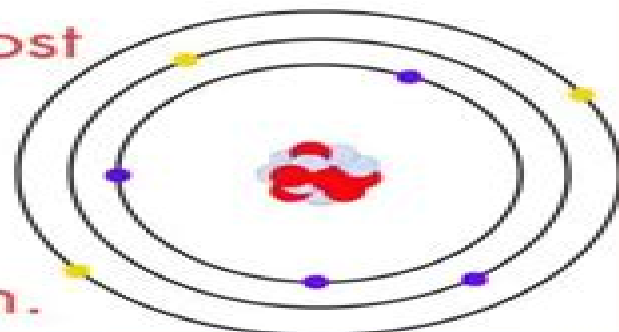
Answer:

- 1st shell = 2 electrons
- 2nd shell = 8 electrons
- 3rd shell = 18 electrons
- 4th shell = 32 electrons (and so on)

- What is a "valence shell"?

Answer:

The valence shell is the outermost electron shell of an atom. The number of electrons in the valence shell determines the chemical properties of the atom.



Electron Arrangements 4 2 Worksheet Answers

Eric Greenberg



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