# **ELECTRONS IN ATOMS**

#### SECTION 5.1 MODELS OF THE ATOM (pages 127-132)

This section summarizes the development of atomic theory. It also explains the significance of quantized energies of electrons as they relate to the quannem mechanical model of the atom.

#### ➤ The Development of Atomic Models (pages 127-128)

 Complete the table about atomic models and the scientists who developed. discorn.

Scientist	Model of Atem
Disition	The atom is a solid indivisible mass.
Thomson	The atom is a ball of positive charge with electrons stuck into the ball.
Rutherford	Most of an atom's mass is concentrated in the small, positively charged nucleus. The electrons surround the nucleus and the rest of the atom is empty space.
Bohr	Electrons are arranged in concentric circular paths around the nucleus,

2. Is the following sentence true or false? The electrons in an atom can exist between energy levels. \_

#### The Bohr Model (pages 128-129)

3. What is a small, discrete unit of energy called?

It is called a quentum.

- 4. Circle the letter of the term that completes the sentence correctly. A quantum of energy is the amount of energy required to
  - a. move an electron from its present energy level to the next lower one
  - b. maintain an electron in its present energy level
- (c.) move an electron from its present energy level to the next higher one
- 5. In general, the higher the electron is on the energy ladder, the

Continue it is from the nucleus.

# Electron Arrangement In Atoms Guided Reading Answers

Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk

#### **Electron Arrangement In Atoms Guided Reading Answers:**

Prentice Hall Chemistry, 2000 AQA AS Chemistry: Unit 1 Foundation Chemistry Student Unit Guide Ebook Margaret Cross, 2008-09-26 Student Unit Guides are perfect for revision Each guide is written by an examiner and explains the unit requirements summarises the relevant unit content and includes a series of specimen questions and answers There are three sections to each guide Introduction includes advice on how to use the guide an explanation of the skills being tested by the assessment objectives an outline of the unit or module and depending on the unit suggestions for how to revise effectively and prepare for the examination questions Content Guidance provides an examiner's overview of the module s key terms and concepts and identifies opportunities to exhibit the skills required by the unit It is designed to help students to structure their revision and make them aware of the concepts they need to understand the exam and how they might analyse and evaluate topics Question and Answers sample questions and with graded answers which have been carefully written to reflect the style of the unit All responses are accompanied by commentaries which highlight their respective strengths and weaknesses giving students an insight into the mind of the examiner Addison-Wesley Chemistry Antony C. Wilbraham, Dennis D. Staley, Michael S. Matta, Edward L. Waterman, Prentice-Hall Staff, 2001-02 To purchase or download a workbook click on the Purchase or Download button to the left To purchase a workbook enter the desired quantity and click Add to Cart To download a free workbook right click the FREE Download PDF link and save to your computer This will result in a faster download as opposed to left clicking and opening the link Fundamentals of Chemistry Burns A. Ralph, Charles Baker, Claire Baker, 1995 This book is a supplement to the texts not a replacement It is intended to maximize your success in this course by showing you how to become involved in developing your own techniques for grasping the concepts of chemistry Using a study outline problem sets problem examples worked and unworked and numerous self tests with answers this manual will provide you with opportunities to sharpen your skills and evaluate your comprehension of the material in the texts The worked out solutions at the end of this manual walk you step by step through the methods of arriving at the answer to those same problems which have an answer only in the answer key Used in conjunction with your textbook and classroom lecture notes this book offers an essential learning opportunity to the chemistry student Back cover Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first

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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

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