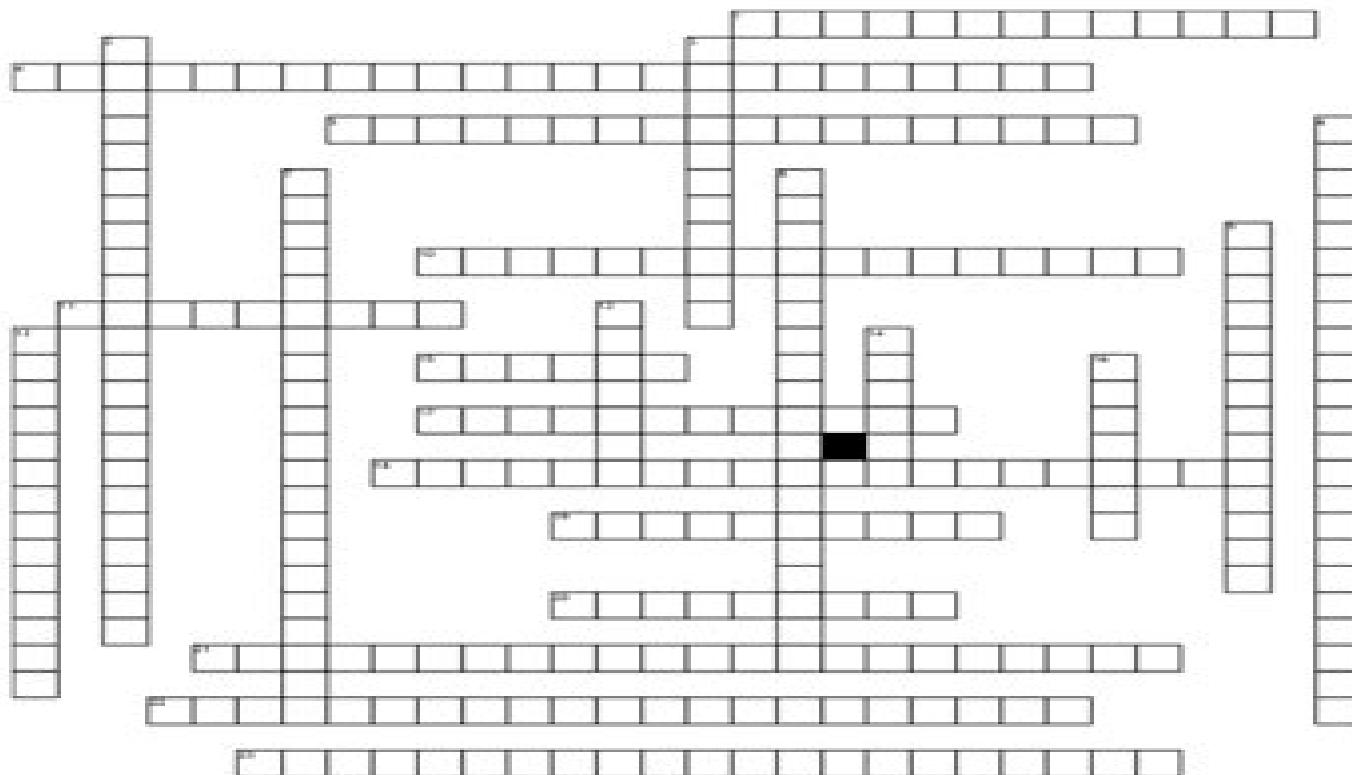


Ch. 4- Arrangement of Electrons in Atoms



Across

1. Describes mathematically the wave properties of electrons and other very small particles.
4. A form of energy that exhibits wavelike behavior as it travels through space.
5. The emission of a continuous range of frequencies of electromagnetic radiation.
10. Indicates the two fundamental spin states of an electron in an orbital.
11. Orbitals of equal energy are each occupied by one electron before any orbital is occupied by a second electron, and all electrons in singly occupied orbitals must have the same spin state.
15. An electron occupies the lowest-energy orbital that can receive it in stated in the _____ principle.
17. The state in which an atom has a higher potential energy than it has in its ground state.

18. When a narrow beam of the emitted light is shined through a prism it is separated into four specific colors of the visible spectrum.

19. The distance between corresponding points on adjacent waves.
20. The number of waves that pass a given point in a specific time.
21. Indicates the main energy level occupied by the electron.
22. Indicates the orientation of an orbital around the nucleus.
23. Refers to an outer main energy level occupied, in most cases by eight electrons.

Down

2. States that it is impossible to determine simultaneously both the position and the velocity of an electron or any other particle.
3. The lowest energy state of an atom.
6. All the forms of electromagnetic radiation for the

7. The arrangement of electrons in an atom is called the _____.

8. The emission of electrons from a metal when light shines on the metal.

9. Specify the properties of atomic orbitals and the properties of the electrons in the orbitals.

12. The minimum quantity of energy that can be lost or gained by an atom.

13. The _____ principle states the no two electrons in the same atom can have the same set of four quantum numbers.

14. A particle of electromagnetic radiation having zero mass and carrying a quantum of energy.

16. A three dimensional region around the nucleus that indicates the probable location of an electron.

Electron Arrangement Crossword Modern Chemistry

Yicheng Fang



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