

Answer Key

PG-1: What is the coefficient of the hydroxide ion for the half reaction when the half-reaction is balanced in basic solution? (Note the coefficient and balanced equation.)



144: 1 (C) 0 (B) 2 (A) 3 (D)



PG-2: What is the balanced half reaction of the species in reacting to form nitrogen monoxide in acidic solution?



PG-3: What is the coefficient for water when the reaction is balanced in acidic solution? (Note the coefficient and balanced equation.)

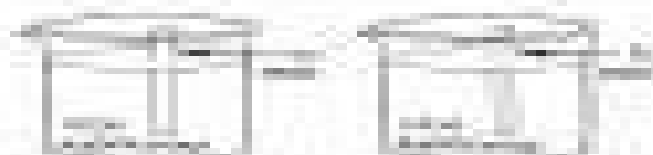


147: 1 (C) 2 (B) 3 (A) 4 (D)



PG-4: Which reaction occurs in the cell in the presence of standard state conditions?

The standard reduction potential of zinc is  $+1.26 \text{ V}$ .  
The standard reduction potential of iron is  $+0.77 \text{ V}$ .



The cell is a galvanic cell. The cell potential is  $+0.49 \text{ V}$ . The cell reaction is  $\text{Zn}(\text{s}) + \text{Fe}^{2+}(\text{aq}) \rightarrow \text{Zn}^{2+}(\text{aq}) + \text{Fe}(\text{s})$ . The cell potential is  $+0.49 \text{ V}$ . The cell reaction is  $\text{Zn}(\text{s}) + \text{Fe}^{2+}(\text{aq}) \rightarrow \text{Zn}^{2+}(\text{aq}) + \text{Fe}(\text{s})$ .

For a reaction, the cell potential and the cell reaction are  $+0.49 \text{ V}$  and  $\text{Zn}(\text{s}) + \text{Fe}^{2+}(\text{aq}) \rightarrow \text{Zn}^{2+}(\text{aq}) + \text{Fe}(\text{s})$ . The cell potential is  $+0.49 \text{ V}$ . The cell reaction is  $\text{Zn}(\text{s}) + \text{Fe}^{2+}(\text{aq}) \rightarrow \text{Zn}^{2+}(\text{aq}) + \text{Fe}(\text{s})$ .

# Electrochemistry Answers Study Guide For Content Mastery

**Sarah Glaven, Uwe Schröder, Eileen  
Hao Yu, Srikanth Mutnuri, Jeffrey A.  
Gralnick, Chi Ho Chan**

## **Electrochemistry Answers Study Guide For Content Mastery:**

**International Society for Microbial Electrochemistry and Technology: Outputs From the 2018 Regional Meetings** Sarah Glaven,Uwe Schröder,Eileen Hao Yu,Srikanth Mutnuri,Jeffrey A. Gralnick,Chi Ho Chan,2020-07-09

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Ponnadurai Ramasami, 2022-10-03 Chapters collected from The Virtual Conference on Chemistry and its Applications VCCA 2021 Research and Innovations in Chemical Sciences Paving the Way Forward This conference was held in August 2021 and organized by the Computational Chemistry Group of the University of Mauritius These peer reviewed chapters offer insights into research on fundamental and applied chemistry with interdisciplinary subject matter

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**How to Survive (and Even Excel In) General Chemistry** Elizabeth Kean, Catherine Middlecamp, 1994 A different kind of book about chemistry which teaches readers the process of learning chemistry not the topic itself Proving a valuable supplement to any introductory text this guide offers inside information to help make chemistry less stressful even enjoyable Includes exercises and sections for self assessment

**The Publisher** , 1906

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