

Problem 13.36 - Enhanced - with Feedback

Ammonia is produced by reacting nitrogen gas and hydrogen gas in the following reaction:



For each of the following changes at equilibrium, indicate whether the equilibrium shifts in the direction of products, shifts in the direction of reactants, or does not change.

Drag the appropriate items to their respective bins.

Reset Help

Shifts equilibrium in direction of reactants

increase the volume of the container

add more $\text{H}_2(\text{g})$

Shifts equilibrium in direction of products

add more $\text{NH}_3(\text{g})$

remove some $\text{N}_2(\text{g})$

Equilibrium does not change

decrease the temperature

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