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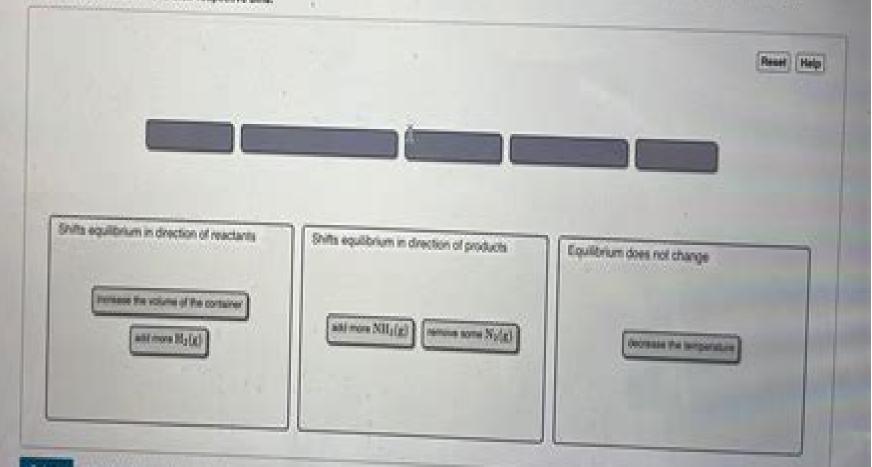
# Problem 13.36 - Enhanced - with Feedback

Ammonia is produced by reacting nitrogen gas and hydrogen gas is the following reaction:

$$N_1(g) + 3H_2(g) := 2NH_2(g) + 92 \; kJ$$

For each of the following changes at equilibrium, indicate whether the equilibrium shifts in the direction of products, shifts in the direction of reactants, or does not change.

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**Léo Pomar** 

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