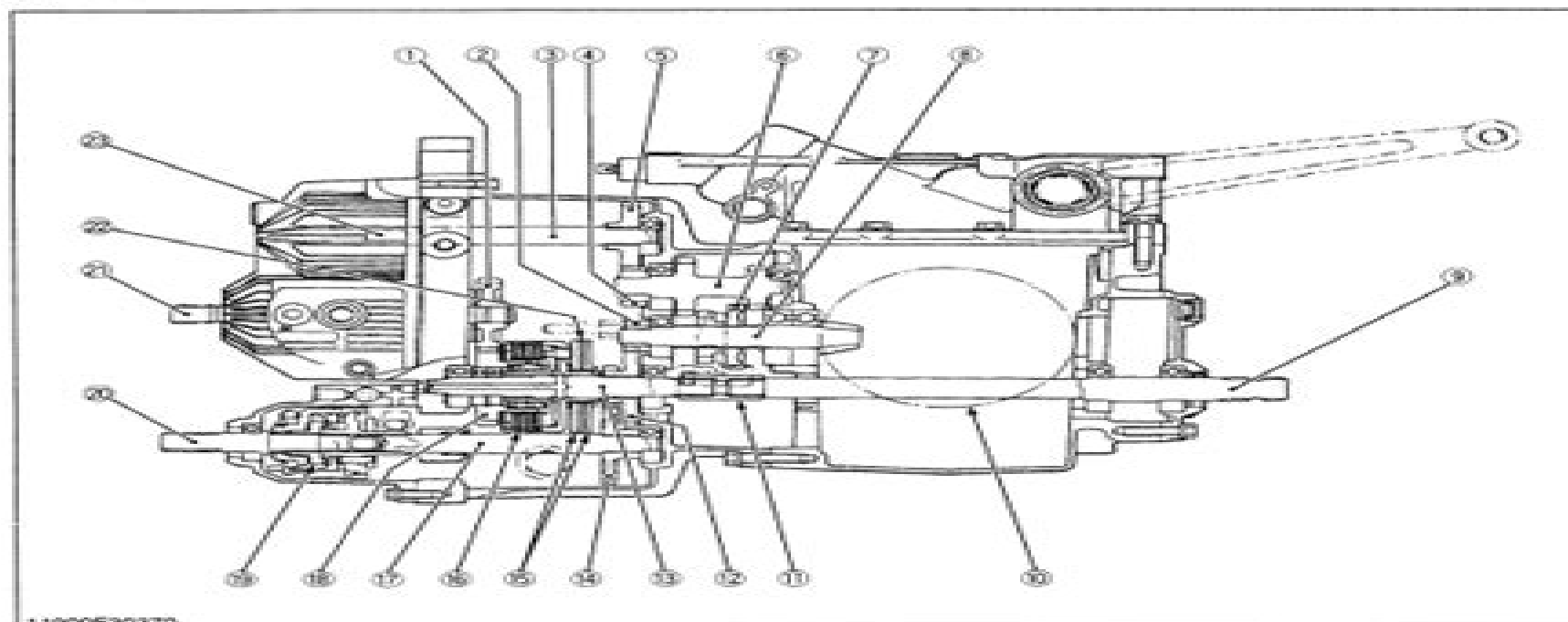


[1] POWER TRAIN

(1) Four Wheel Drive



11900F30370

- | | | | |
|------------------|-------------------------------|--|-------------------------------------|
| (1) 30T Gear | (8) Spiral Bevel Pinion Shaft | (15) PTO Brake Plate | (20) RWD Shaft 2 |
| (2) 11T Gear | (9) PTO Shaft | (16) Hydraulic PTO Clutch Assembly | (21) Pump Shaft |
| (3) Motor Shaft | (10) Differential Gear | (17) RWD Shaft 1 | (22) PTO Brake Disc |
| (4) 15T Gear | (11) Coupling | (18) 32T Gear | (23) Hydrostatic Transmission (HST) |
| (5) 28T Gear | (12) 16T-33T Gear | (19) Dual-acting Overrunning Clutch (DAO Clutch) | |
| (6) 18T-17T Gear | (13) Clutch Shaft | | |
| (7) 15T-26T Gear | (14) 34T Gear | | |

The transmission consists of a series of gears shown above. It offers the most suitable speed for travelling and operation by combination of these gears.

It transmits power to the front and rear axles and PTO shaft, which are classified respectively as the travelling system and PTO system.

FORWARD AND REVERSE

■ Hi Position

Motor Shaft (3) → 28T Gear (5) → 15T Gear (4) → 18T-(17T) Gear (6) → 15T-(26T) Gear (7) → Spiral Bevel Pinion Shaft (8) → Differential Gear (10)

■ Low Position

Motor Shaft (3) → 28T Gear (5) → 15T Gear (4) → (18T)-17T Gear (6) → (15T)-26T Gear (7) → Spiral Bevel Pinion Shaft (8) → Differential Gear (10)

■ Neutral Position

Motor Shaft (3) → 28T Gear (5) → 15T Gear (4) → 18T-17T Gear (6)

■ Rear Wheel Drive

Spiral Bevel Pinion Shaft (8) → 11T Gear (2) → 16T-33T Gear (12) → RWD Shaft 1 (17) → DAO Clutch (19) → RWD Shaft 2 → Rear Differential Gear

11900MC30010

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Robert R. Redfield



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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414 l 22 414 l we can substitute 101 325 kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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introduction the ideal gas law is a commonly used formula in chemistry it relates pressure temperature and volume this
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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of
declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95
82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in
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condition at which a phase change occurs from liquid to vapour or vapour to liquid

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the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates
the volume and the number of particles in each flask indicates relative amounts