# Practice: Balancing Equations #3

Show all work on a separate sheet of paper. Place coefficient final answers in the blanks provided. You can leave

<sup>2</sup>. \_\_\_\_MgF<sub>2</sub>+ \_\_\_\_Li<sub>2</sub>CO<sub>3</sub> 
$$\Rightarrow$$
 \_\_\_\_MgCO<sub>3</sub>+ \_\_\_\_LiF

3. \_\_\_ 
$$P_4$$
 + \_\_\_  $O_2$   $\Rightarrow$  \_\_\_  $P_2O_3$ 

4. \_\_\_\_ 
$$RbNO_3$$
 + \_\_\_\_  $BeF_2$   $\rightarrow$  \_\_\_\_  $Be(NO_3)_2$  + \_\_\_\_  $RbF$ 

5. \_\_\_\_ AgNO<sub>3</sub> + \_\_\_\_ Cu 
$$\rightarrow$$
 \_\_\_\_ Cu(NO<sub>3</sub>)<sub>2</sub> + \_\_\_\_ Ag

6. \_\_\_\_ 
$$CF_4 +$$
 \_\_\_\_  $Br_2 \Rightarrow$  \_\_\_\_  $CBr_4 +$  \_\_\_\_  $F_2$ 

7. \_\_\_\_ HCN + \_\_\_ CuSO<sub>4</sub> 
$$\Rightarrow$$
 \_\_\_\_ H<sub>2</sub>SO<sub>4</sub> + \_\_\_ Cu(CN)<sub>2</sub>

$$10.$$
\_\_\_N<sub>2</sub>+\_\_\_NH<sub>3</sub>

11. NaF + Br<sub>2</sub> 
$$\rightarrow$$
 NaBr + F<sub>2</sub>

12. 
$$Pb(OH)_2 + HCI \rightarrow H_2O + PbCl_2$$

13.\_\_\_\_ 
$$AlBr_3 +$$
 \_\_\_\_  $K_2SO_4 \rightarrow$  \_\_\_\_  $KBr +$  \_\_\_\_  $Al_2(SO_4)_3$ 

14.\_\_\_ 
$$CH_4+$$
 \_\_\_  $O_2 \rightarrow$  \_\_\_  $CO_2+$  \_\_\_  $H_2O$ 

15.\_\_\_\_ Na<sub>3</sub>PO<sub>4</sub>+\_\_\_\_ CaCl<sub>2</sub>
$$\rightarrow$$
 \_\_\_\_ NaCl + \_\_\_\_ Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>

17. 
$$AI + HCI \rightarrow H_2 + AICI_3$$

18.\_\_\_ 
$$N_2 +$$
\_\_\_  $F_2 \rightarrow$ \_\_\_  $NF_3$ 

19.\_\_\_ 
$$SO_2 +$$
 \_\_\_  $Li_2Se \rightarrow$  \_\_\_  $SSe_2 +$  \_\_\_  $Li_2O$ 

20. 
$$NH_3 + H_2SO_4 \Rightarrow (NH_4)_2SO_4$$

# **How To Balance Equations Grade 9 Ns**

**K Morrison** 

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