Answer kev-H.W(2)-Oxidation-reduction reaction

1-Multiple choice guestions:

1-Oxidation involves

a-gaining electrons; gaining oxygen b-losina electrons: losina oxvaen c-aaining electrons; losing oxygen d-losing electrons; gaining oxygen

2- Reduction involves

a-gaining electrons; gaining oxygen b-losing electrons; losing oxygen c-agining electrons, losing oxygen d-losing electrons; gaining oxygen

3-In the reaction

2Ca(s) + O2(g) --> 2CaO(s) calcium is _ a-Reduced b- Synthesized

4- In the reaction Zn + H2O --> ZnO2 + H2 which element, is oxidized?

a-Zinc b- Hydrogen c- Oxygen

5- Which substance is reduced in the following reaction?

NaOH + Li --> LiOH + Na

42-41 b-Mar C-O4-14

6-Which of the following is the balanced half -reaction for the oxidation of Cu to Cu-27

a-Cu -> Cu*2 b-Cu + 2 e- -> Cu*2 c-Cu -> Cu*2 + 2 e-

7-Another name for a "oxidation-reduction" reaction is

a-chemical reaction b-neutralization reaction c-redox reaction

Remember

Redox (Oxidation and Reduction)

Oxidation	Reduction
Gain oxygen †	Loss oxygen
Gain oxidation Number †	Loss exidation Number [
Loss hydrogen;	Gain hydrogen

c-Oxidized

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Marco Cascella

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