Directions: Match each term in Column I with its description in Column II. Write the letter of the correct term in the space provided.

## Column I binary compound chemically stable 3. nonpolar molecule \_\_\_\_\_ 4. ion \_\_\_\_ 5. ionic bond 6. polar molecule 7. oxidation number 8. chemical formula 9. covalent bond \_\_\_\_\_ 10. hydrate ----- 11. chemical bond \_\_\_\_\_ 12. polyatomic ion

## Column II

a. number that indicates how many electrons an atom must gain, lose, or share to become stable

Class

- b. shorthand that tells what elements a compound contains and the exact number of atoms of each element in a unit of the compound
- c. positively or negatively charged, covalently bonded group of atoms
- compound composed of two elements
- e. describes an atom that has a full outermost energy level
- f. molecule that has a slightly positive end and a slightly negative end
- g. the attraction that forms between atoms when they share electrons
- h. the force that holds atoms together in a compound
- a compound that has water chemically attached to it
- j. the force of attraction between the opposite charges of the ions in an ionic compound
- k. molecule made of two identical atoms that share the electrons equally
- a charged particle that has either more or fewer electrons than protons

# **Gpscience Physical Science Chemical Bonds Answer**

**Zvonimir B. Maksic** 

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Chemical Bonds Phillip Manning, 2009 Modern life is made up of a mind boggling array of materials A simple drinking cup for example might be made of Styrofoam paper or glass depending on the drinkers needs at the moment Home storage cabinets can be made of metal wood or plastic Space shuttles are assembled from silicon steel and hundreds of other materials All of these items owe their properties to the chemical bonds between the atoms that make up the substance Chemical Bonds examines the nature of the chemical bonds answering fundamental questions about how they form how they are broken and how they help define life as we know it Chemical Bonding The Open The Open Courses Library, 2019-12-02 Chemical Bonding Grade 10 Physical Science When you look at the matter or physical substances around you you will realise that atoms seldom exist on their own More often the things around us are made up of different atoms that have been joined together This is called chemical bonding Chemical bonding is one of the most important processes in chemistry because it allows all sorts of different molecules and combinations of atoms to form which then make up the objects in the complex world around us Chapter Outline Covalent Bonding Lewis structures Ionic bonding Metallic Bonding Writing formulae The Open Courses Library introduces you to the best Open Source Courses Chemical Bonds and Bonds Energy R Sanderson, 2012-12-02 Chemical Bonds and Bonds Energy Second Edition provides information pertinent to the fundamental aspects of contributing bond energy and bond dissociation energy. This book explores the values that are useful in the interpretation of significant phenomena such as product distribution and reaction mechanisms Organized into 12 chapters this edition begins with an overview of the quantitative relationship among three basic properties of an atom namely nonpolar covalent radius electronegativity and homonuclear single covalent bond energy This text then examines the quantitative means of evaluating the partial atomic charges that result from initial differences in the electromagnetivity of atoms that form a compound Other chapters consider the recognition of the reduction of bond weakening not by multiplicity and in certain types of single covalent bonds The final chapter deals with the application of the principal ideas and techniques to the oxidation of ethane This book is a valuable resource for organic and inorganic chemists The Chemical Bond Gernot Frenking, Sason Shaik, 2014-06-13 This is the perfect complement to Chemical Bonding Across the Periodic Table by the same editors who are two of the top scientists working on this topic each with extensive experience and important connections within the community The resulting book is a unique overview of the different approaches used for describing a chemical bond including molecular orbital based valence bond based ELF AIM and density functional based methods It takes into account the many developments that have taken place in the field over the past few decades due to the rapid advances in quantum chemical models and faster computers Chemical Bonds Jeremy K. Burdett, 1997-05-28 Inorganic Chemistry This series reflects the breadth of modern research in inorganic chemistry and fulfils the need for advanced texts The series covers the whole range of inorganic and physical chemistry solid state chemistry coordination

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Chemical Bonds and Bond Energy Robert Thomas Sanderson,1971 Linus Pauling - Selected Scientific Papers (In 2 Volumes) - Volume 1 Barclay Kamb, Linda Pauling Kamb, Carl Alexander Kamb, Peter Jeffress Pauling, Linus Pauling Jr,2001-11-02 Linus Pauling wrote a stellar series of over 800 scientific papers spanning an amazing range of fields some of which he himself initiated This book is a selection of the most important of his writings in the fields of quantum mechanics chemical bonding covalent ionic metallic and hydrogen bonding molecular rotation and entropy protein structure hemoglobin molecular disease molecular evolution the antibody mechanism the molecular basis of anesthesia orthomolecular medicine radiation chemistry biology and nuclear structure Through these papers the reader gets a fresh unfiltered view of the genius of Pauling s many contributions to chemistry chemical physics molecular biology and molecular medicine Chemical Bonds in Semiconductors and Solids N. N. Sirota, 1995-12-31 Chemical Bonds in Solids N.N. Sirota, 2014-01-17

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