

General Science **Part-11**

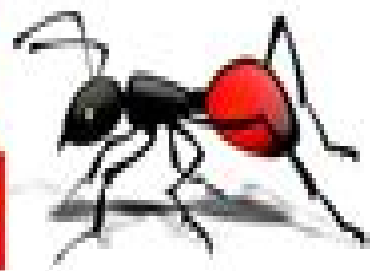
Q. Which gas is used in aeroplane tyres?

- (A) Nitrogen (B) Helium
(C) Neon (D) Hydrogen



Q. When an ant bite us, they inject

- (A) Formic acid (B) Malic acid
(C) Tartaric acid (D) Acetic acid



Material Science Multiple Choice Question Answer

Didier Musso



Material Science Multiple Choice Question Answer:

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web donor whereas a base generically represented by b is a proton acceptor when acids and bases combine they swap a proton ha b a hb strong acids e g sulfuric acid have a very pronounced tendency to transfer a proton to a base whereas weak acids

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web acid base titrations laboratory report name rj hamdan date 11 8 section 432 ta ben kies a goals purpose of experiment 2 points 50 words the goal of this experiment is using the skills of precision to determine the

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web understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web jul 19 2023 at the beginning of the titration the solution in the erlenmeyer flask is acidic as the base is added it completely reacts with the acid and the solution in the erlenmeyer flask continues to be acidic but at the equivalence point

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web answer based on the data the strong acid strong base combination of hydrochloric acid and sodium hydroxide had the neutral pH at the equivalence point being 7 this result is expected because a combination of strong acids and strong bases should result in a neutral at the equivalence point

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web 1 approximately 3 g of hydrated oxalic acid $H_2C_2O_4 \cdot 2H_2O$ was weighted to the nearest 0.01 g in a 50 mL beaker 2 about

30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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web solution a titrant volume 0 ml the solution pH is due to the acid ionization of HCl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the pH of the solution is then $\text{pH} = -\log 0.100 = 1.000$ pH $\log 0.100 = 1.000$ b titrant volume 12.50 ml

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to determine the concentration of an acid with a base to determine the concentration of an acidic sample

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