

Significant Figure and Scientific Notation Worksheet

1. Convert each of the following into scientific notation.

- | | |
|-------------------------|---------------------------|
| a) 3427 | k) 2205.2 |
| b) 0.00456 | l) 30.0×10^{-2} |
| c) 123,453 | m) 0.982×10^{-3} |
| d) 172 | n) 0.0473 |
| e) 0.000984 | o) 650,502 |
| f) 0.502 | p) 3.03×10^{-1} |
| g) 3100.0×10^2 | q) 20.4×10^5 |
| h) 0.0114×10^4 | r) 1000×10^{-3} |
| i) 107.2 | |
| j) 0.0000455 | |

2. Determine the number of significant digits in each of the following:

- | | | |
|-------------|--------------------------|---------------------------|
| a) 3427 | g) 3100.0×10^2 | m) 0.982×10^{-3} |
| b) 0.00456 | h) 0.0114×10^4 | n) 0.0473 |
| c) 123,453 | i) 107.2 | o) 650,502 |
| d) 172 | j) 0.0000455 | p) 3.03×10^{-1} |
| e) 0.000984 | k) 2205.2 | q) 20.4×10^5 |
| f) 0.502 | l) 30.0×10^{-2} | r) 1000×10^{-3} |

3. Convert each into decimal form.

- | | |
|---------------------------|----------------------|
| a) 1.56×10^4 | g) 6.9×10^4 |
| b) 0.56×10^{-2} | |
| c) 3.69×10^{-2} | |
| d) 736.9×10^5 | |
| e) 0.00259×10^5 | |
| f) 13.69×10^{-2} | |

Mastering Chemistry Significant Figure Answers

Mary L. Kotz, Alton J. Banks



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