
Maximum Marks: 80

Time allowed Two hours

Answers to this Paper must be written on the paper provided separately

You will not be allowed to write during first 15 minutes.

This time is to be spent in reading the question paper.

The time given at the head of this Paper is the time allowed for writing the answers.

Section A is compulsory. Attempt any four questions from Section B.

The intended marks for questions or parts of questions are given in brackets [].

SECTION A (40 Marks)

(Attempt all questions from this Section.)

Question 1

Choose the correct answers to the questions from the given options.

[15]

(Do not copy the questions, write the correct answers only.)

(i) An element in period 3, whose electron affinity is zero:

- (a) Neon
- (b) Sulphur
- (c) Sodium
- (d) Argon

Answer: (d)

Argon belongs to period 3. It is an inert element and stable. It neither requires gain nor loss. It has its octet complete and hence it has the zero electron affinity.

(ii) An element with the largest atomic radius among the following is:

- (a) Carbon
- (b) Nitrogen

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**Thirumalaisamy P.
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Table of Contents Grade 10 Chemistry Exam Papers And Memos

1. Understanding the eBook Grade 10 Chemistry Exam Papers And Memos
 - The Rise of Digital Reading Grade 10 Chemistry Exam Papers And Memos
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 10 Chemistry Exam Papers And Memos
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 10 Chemistry Exam Papers And Memos
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 10 Chemistry Exam Papers And Memos
 - Personalized Recommendations
 - Grade 10 Chemistry Exam Papers And Memos User Reviews and Ratings
 - Grade 10 Chemistry Exam Papers And Memos and Bestseller Lists
5. Accessing Grade 10 Chemistry Exam Papers And Memos Free and Paid eBooks
 - Grade 10 Chemistry Exam Papers And Memos Public Domain eBooks
 - Grade 10 Chemistry Exam Papers And Memos eBook Subscription Services
 - Grade 10 Chemistry Exam Papers And Memos Budget-Friendly Options
6. Navigating Grade 10 Chemistry Exam Papers And Memos eBook Formats

- ePub, PDF, MOBI, and More
 - Grade 10 Chemistry Exam Papers And Memos Compatibility with Devices
 - Grade 10 Chemistry Exam Papers And Memos Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 10 Chemistry Exam Papers And Memos
 - Highlighting and Note-Taking Grade 10 Chemistry Exam Papers And Memos
 - Interactive Elements Grade 10 Chemistry Exam Papers And Memos
 8. Staying Engaged with Grade 10 Chemistry Exam Papers And Memos
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 10 Chemistry Exam Papers And Memos
 9. Balancing eBooks and Physical Books Grade 10 Chemistry Exam Papers And Memos
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 10 Chemistry Exam Papers And Memos
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Grade 10 Chemistry Exam Papers And Memos
 - Setting Reading Goals Grade 10 Chemistry Exam Papers And Memos
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Grade 10 Chemistry Exam Papers And Memos
 - Fact-Checking eBook Content of Grade 10 Chemistry Exam Papers And Memos
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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