

- Silver reacts with H_2S present in the atmosphere to form a black color compound Silver Sulphide. This phenomenon is called as corrosion.
- Black color compound formed is Silver Sulphide



Long Answer Questions

(a) Write a balanced chemical equation of the reaction.

(b) Identify the brown gas X evolved.

(c) Identify the type of reaction.

(d) What could be the pH range of aqueous solution of the gas X?

100



(b) The brown gas X is nitrogen dioxide, NO_2 .

(c) Reaction involved is Thermal decomposition.

(d) $\text{pH} < 7$ because NO_2 dissolves in water to form acidic solution (pH range below 7)

40. Give the characteristic tests for the following gases

(m) CO₂

202

161 03

Webb H2

Sediment

a) Pass CO_2 into limewater which will turn water into milky. This is the confirmation test for the presence of Carbon-di-oxide.

b) Smell is the characteristic feature of SO_2 which smells like rotten egg.

c) Test for oxygen involves burning of match stick near oxygen makes it burn even more brightly.

d) When Burning matchstick is brought near H_2 gas, the flame burns with the pop sound. This is the test to confirm Hydrogen gas.

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