

Chemistry – Mock Test Paper

[Time: $1\frac{1}{2}$ hrs]

[M. Marks : 80]

Answers to this paper must be written on the paper provided separately.

You will not be allowed to write during the first 15 minutes.

This time is to be spent in reading the Question Paper.

The time given at the head of this paper is the time allowed for writing the answers.

Section I is compulsory. Attempt any four questions from Section II.

Section I (40 Marks)

Attempt all questions from this section

Question 1.

(a) Choose the most appropriate answer.

- (i) The element with largest size in second period is :
 A. Lithium B. Oxygen C. Fluorine D. Carbon
- (ii) Which is the element which belongs to group 13 from :
 A. Beryllium B. Boron C. Aluminium D. Carbon
- (iii) The most strong alkali is
 A. Lithium hydroxide B. Potassium hydroxide
 C. Sodium hydroxide D. Rubidium hydroxide
- (iv) Lanthanides is a group of elements outside the Periodic Table having atomic number from
 A. 57 to 71 B. 89 to 103 C. 72 to 86 D. 104 to 118
- (v) The elements with lowest first ionisation energy is
 A. Sodium B. Caesium C. Barium D. Magnesium
- (vi) When equal volumes of hydrogen gas and chlorine gas are exposed to diffused sunlight the reaction :
 A. does not take place B. takes place at moderate speed
 C. is explosive in nature D. none of the above.
- (vii) In laboratory ammonia gas prepared by heating a mixture of :
 A. NH_4Cl and $\text{Ca}(\text{OH})_2$ B. $(\text{NH}_4)_2\text{SO}_4$ and $\text{Ca}(\text{OH})_2$
 C. NH_4NO_3 and KOH D. NH_4NO_3 and $\text{Ca}(\text{OH})_2$
- (viii) The compound which will liberate carbon dioxide on treating with dilute sulphuric acid is :
 A. MgCl_2 B. Na_2SO_4 C. Na_2CO_3 D. Na_2SO_4
- (ix) Which is not a member of alkanolic series?
 A. HCOOH B. $\text{C}_2\text{H}_5\text{COOH}$ C. HCHO D. CH_3COOH
- (x) Ethyne can be converted into ethyne ozonide by treating with :
 A. Hydrogen peroxide B. Ozone
 C. Oxygen D. Conc. HNO_3

[10]

Grade 10 Chemistry Exam Papers

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Grade 10 Chemistry Exam Papers:

The book delves into Grade 10 Chemistry Exam Papers. Grade 10 Chemistry Exam Papers is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Grade 10 Chemistry Exam Papers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Grade 10 Chemistry Exam Papers
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- Chapter 3: Grade 10 Chemistry Exam Papers in Everyday Life
- Chapter 4: Grade 10 Chemistry Exam Papers in Specific Contexts
- Chapter 5: Conclusion

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