

Chemical Reactions

Different Ways of Representing Chemical Reactions:

1. Word Equation - a description of a chemical reaction using words
 i.e. sodium + chlorine $\rightarrow$ sodium chloride
 magnesium + oxygen $\rightarrow$ magnesium oxide

2. Skeleton Equation - a description of a chemical reaction using chemical formulas
 i.e. $\text{Na(s)} + \text{Cl}_2(\text{g}) \rightarrow \text{NaCl(s)}$
 $\text{Mg(s)} + \text{MgCl}_2(\text{aq}) \rightarrow \text{Mg(s)} + \text{NaCl(aq)}$

**diatomic $\rightarrow 2 = 1$ **

include states for each substance (l) (s) (g) (aq)

3. Balanced Chemical Equations - a description of a chemical reaction using chemical formulas and co-efficients to ensure that the number of atoms of each elements is equal for reactants and products

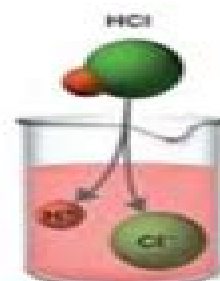
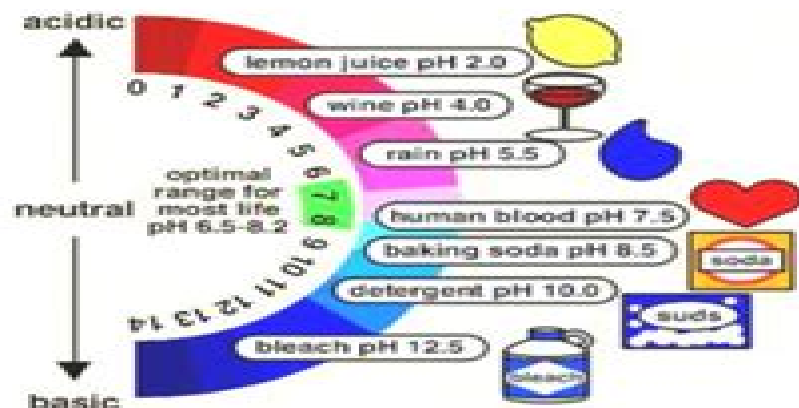


include states for each substance (l) (s) (g) (aq)

Acids & Bases:

Acids - contain a hydrogen ion

Bases - contain a hydroxide ion



(a) Acid



(b) Base



(c) Salt

Gr 11 Chem Reactions Quiz

Scott C. Dulebohn

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