

Name _____ Date _____ Per _____

Molecular Formula

- = empirical formula _____ number of _____ of each
in this _____
- = not necessary for _____
- = necessary for _____

The molecular formula for water is _____ and empirical formula is also

The molecular formula for hydrogen peroxide is _____ and empirical
formula is _____

Example Problem

The empirical formula for glucose is CH_2O

a) If the molar mass is 180.12 g/mol , find the molecular formula.

b) If the molar mass is _____ g/mol , find the molecular formula.

Problem Set One (write answers on paper)

empirical formula	molar mass
CH_2	_____ g/mol
FCl_2	_____ g/mol
C_2H_4	_____ g/mol

Ex. Problem: Find the molecular formula for a compound with
_____ g H _____ g O molar mass _____ g/mol

Gpb Chemistry Episode 703 Answers

Yan Bai



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