

## Substituted Molecular Compounds

Name the following using Substituted Compounds

Take great care that Compounds that are substituted don't have

Number

in their compounds, which are instead known as compounds with substituents. The atoms bond together to form compounds with substituents.

Compounds with substituents are used to show the number of atoms of each element per molecule. This is shown below. Using this prefix and the number of atoms each represents:

mono	di	tri	tetra	penta	hexa	hepta	octa	nona	deca
1	2	3	4	5	6	7	8	9	10

Write the formula of these substituted compounds:

carbon tetrachloride  $\text{CCl}_4$

sulfur dioxide  $\text{SO}_2$

dinitrogen monoxide

$\text{N}_2\text{O}$

dinitrogen trioxide  $\text{N}_2\text{O}_3$

dinitrogen pentoxide

$\text{N}_2\text{O}_5$

nitrogen dioxide  $\text{NO}_2$

phosphorus trichloride  $\text{PCl}_3$

carbon disulfide  $\text{CS}_2$

triphosphorus dioxide  $\text{P}_3\text{O}_2$

carbon monoxide  $\text{CO}$

## Organic Compounds in General Cases of Molecular Compounds

Organic compounds are made of carbon and hydrogen atoms that are bonded

Carbon. The number of the carbon atoms each unit of

carbon atoms in an organic compound will determine whether the compound is saturated or unsaturated. This is shown below. The number of atoms in a unit is

saturated compound are single bonds. Each unit of unsaturated compound

that has a double bond is unsaturated and double bond.

single bonds. The organic compounds containing only hydrogen and

oxygen are hydrocarbons.

Hydrocarbons

are

# Gpb Chemistry Note Answers

**Anthony S. Fauci**



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