

Chemistry Formulas



Dilution

Energy of a photon

Pressure

Average Kinetic Energy

Ideal Gas Law

Combined Gas Law

Dalton's Law of Partial Pressures

Dalton's Law of Partial Pressures

Graham's Law of Effusion

Henry's Law

Freezing Point Depression

Boiling Point Elevation

Vapor Pressure Depression (Raoult's Law)

Osmotic Pressure (#)

Arrhenius Equation

Equilibrium Constant Expressions

Equilibrium Constant Expressions

Equilibrium Constant Expressions

Reaction Quotient (Q)

Solubility Product Constant (K_{sp})

pH & pOH

рн & рон

pH & pOH

 $M_*V_* = M_*V_*$

 $E_{\text{photon}} = hf = hc/\gamma$

P = F/A

 $KE_{max} = \frac{3}{2}RT$

PV = nRT

 $\frac{P_{+}V_{+}}{N_{+}T_{+}} = \frac{P_{+}V_{+}}{N_{+}T_{+}}$

P total = PA + PB+

 $P_A = \chi A P_{\text{total}}$

 $\frac{r_{\perp}}{r_{\perp}} = \sqrt{\frac{M_{\perp}}{M_{\perp}}}$

 $P_A = k_B [A]$

 $\Delta T_F = -iK_F M$

 $dT_B = iK_B m$

 $P_{soln} = \chi_{solv} P^{o}_{solv}$

 $\pi = iMRT$

 $k = Ae^{-\epsilon \omega/\pi \tau}$

 $K_{e} = \frac{[products]}{[reactures]}$

 $K_{eq} = \frac{k_{rooment}}{k_{roomer}}$

 $K_P = \frac{P_{\text{maximize}}}{P_{\text{maximize}}}$

 $Q = \frac{[productets]}{[resetants]}$

 $K_{sp} = \frac{[products]}{[reactants]}$

 $pH = -\log[H^{-\epsilon}]$

 $pOH = -\log[OH]$ pH + pOH = 1+ pK, & pK,

pK, & pK, pK, & pK,

Neutralization Reaction

Buffers

Enthalpy of Formation

First Law of Thermodynamics

Pressure Volume Work

Calorimetry Thermal Energy (q)

Heat Curves & Thermal Energy (g)

Heat Curves & Thermal Energy (q)

Heat Curves & Thermal Energy (q)

Entropy (S)

Bond Dissociation Energy

Gibb's Free Energy (AG)

Gibb's Free Energy (AG)

Gibb's Free Energy (AG)

Standard Cell Potential

Standard Cell Potential

Nernst Equation

Kinetics (always 1st order)

Kinetics (always 1st order)

Kinetics (always 1st order)

Nuclear Binding Energy

 $pK_a = -\log[K_a]$

 $pK_b = -\log[K_b]$

 $pK_a + pK_b = 14$

 $n_A M_A V_A = n_B M_B V_B$

 $pH = pKa + \log [A - 1]HA$

 $\Delta H_f = En\Delta H^*_f(product) -$

EndII (reactants)

 $\Delta E = q + w$

 $W = -P\Delta V$

 $q = -C_{cutorimeter}\Delta T$

 $q = mc\Delta T$

 $q = m\Delta H_{funion}$

 $q = m\Delta H_{papartrustum}$

 $\Delta S = EnS_{products} - EnS_{reactants}$

 $\Delta H = \Sigma \Delta H_{rescence} - \Sigma \Delta H_{products}$

 $\Delta G^* = \Delta H^* - T\Delta S^*$

 $\Delta G = \Delta G^* + RT lnQ$

 $\Delta G^* = -RT^*lm(K_{eq})$

 $E^* = E^*reduction + E^*oxidation$

 $E^* = E^* cathode + E^* anode$

 $E_{cell} = E^* - \frac{0.0592}{5} \log \Omega$

 $N = N_e e^{-kz}$

 $\ln N = \ln N_n - kt$

In N = In No - Kt

 $t_{1/2} = \frac{0.693}{k}$

 $E = A mc^2$

Formula Sheet For Gen Chem Chm 2045

Hussin A.Rothana

Formula Sheet For Gen Chem Chm 2045:

The book delves into Formula Sheet For Gen Chem Chm 2045. Formula Sheet For Gen Chem Chm 2045 is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Formula Sheet For Gen Chem Chm 2045, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
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 - Chapter 2: Essential Elements of Formula Sheet For Gen Chem Chm 2045
 - Chapter 3: Formula Sheet For Gen Chem Chm 2045 in Everyday Life
 - Chapter 4: Formula Sheet For Gen Chem Chm 2045 in Specific Contexts
 - ∘ Chapter 5: Conclusion
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- 3. In chapter 2, the author will delve into the foundational concepts of Formula Sheet For Gen Chem Chm 2045. The second chapter will elucidate the essential principles that need to be understood to grasp Formula Sheet For Gen Chem Chm 2045 in its entirety.
- 4. In chapter 3, the author will examine the practical applications of Formula Sheet For Gen Chem Chm 2045 in daily life. This chapter will showcase real-world examples of how Formula Sheet For Gen Chem Chm 2045 can be effectively utilized in everyday scenarios.
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