

acid) pH 1			hydrogen carbonate ('bicarb', baking soda) pH 8	13-14
HCl hydrochloric acid (in the lab is same as your stomach!) pH 1	CH ₃ COOH acetic/ethanoic acid (vinegar)	NaCl sodium chloride (salt water) pH 7	NH ₃ ammonia pH 11	KOH potassium hydroxide pH 13-14
HNO ₃ nitric acid pH 1	Fruit juices eg oranges and lemons contain citric acid pH 2-3	MgSO ₄ magnesium sulphate (Epsom salts) pH 6.5-7.0	Na ₂ CO ₃ sodium carbonate (washing soda) pH 11	oven cleaner (may contain NaOH) pH can be >12
	rain water (pH 5.5 normal, down to 3.5 when polluted)	C ₆ H ₁₂ O ₆ sugar pH 7	toothpaste pH 8	
	milk pH 6	C ₂ H ₅ OH ethanol (alcohol) pH 7	bleach pH 11	
	wine/beer pH 6		Ca(OH) ₂ calcium hydroxide (limewater) pH 10	
			Mg(OH) ₂ magnesium hydroxide ('milk of magnesia') pH 10	

Form Chemistry Notes

Cecilia Genereux, Paul Moeller



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Lecture Notes in Quantum Chemistry II Björn O. Roos, 2012-12-06 The first volume of Lecture Notes in Quantum Chemistry Lecture Notes in Chemistry 58 Springer Verlag Berlin 1992 contained a compilation of selected lectures given at the two first European Summer Schools in Quantum Chemistry ESQC held in southern Sweden in August 1989 and 1991 respectively The notes were written by the teachers at the school and covered a large range of topics in ab initio quantum chemistry After the third summer school held in 1993 it was decided to put together a second volume with additional material Important lecture material was excluded in the first volume and has now been added Such added topics are integrals and integral derivatives SCF theory coupled cluster theory relativity in quantum chemistry and density functional theory One chapter in the present volume contains the exercise material used at the summer school and in addition solutions to all the exercises It is the hope of the authors that the two volumes will find good use in the scientific community as textbooks for students who are interested in learning more about modern methodology in molecular quantum chemistry The books will be used as teaching material in the European Summer Schools in Quantum Chemistry which are presently planned Lund in July 1994 Bjorn Roos NOTES ON HARTREE FOCK THEORY AND RELATED TOPICS Jan Almlöf Department of Chemistry University of Minnesota Minneapolis MN 55455 USA Contents 1 Introduction 2 The Born Oppenheimer Approximation 3 Determinant Wavefunctions and the Pauli Principle 4 Expectation Values With a Determinant Wavefunction

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